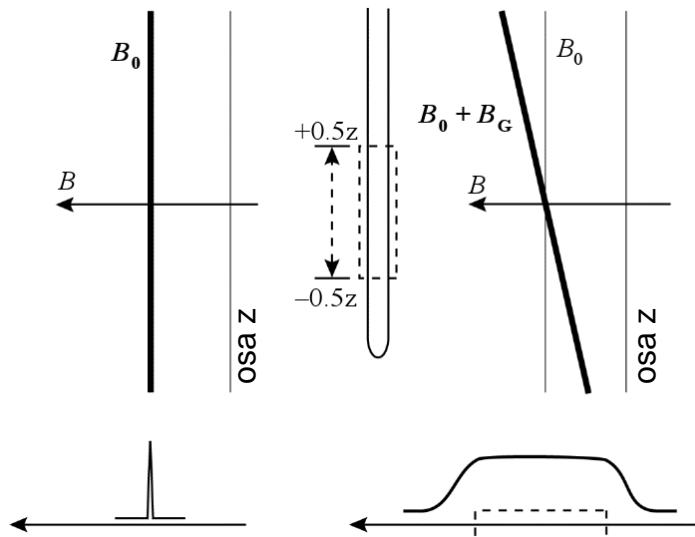


# Gradient magnetického pole



Pomocí dalších cívek zajistíme, aby se magnetické pole  $B_0$  měnilo lineárně v závislosti na poloze

Gradient lze zapínat a vypínat

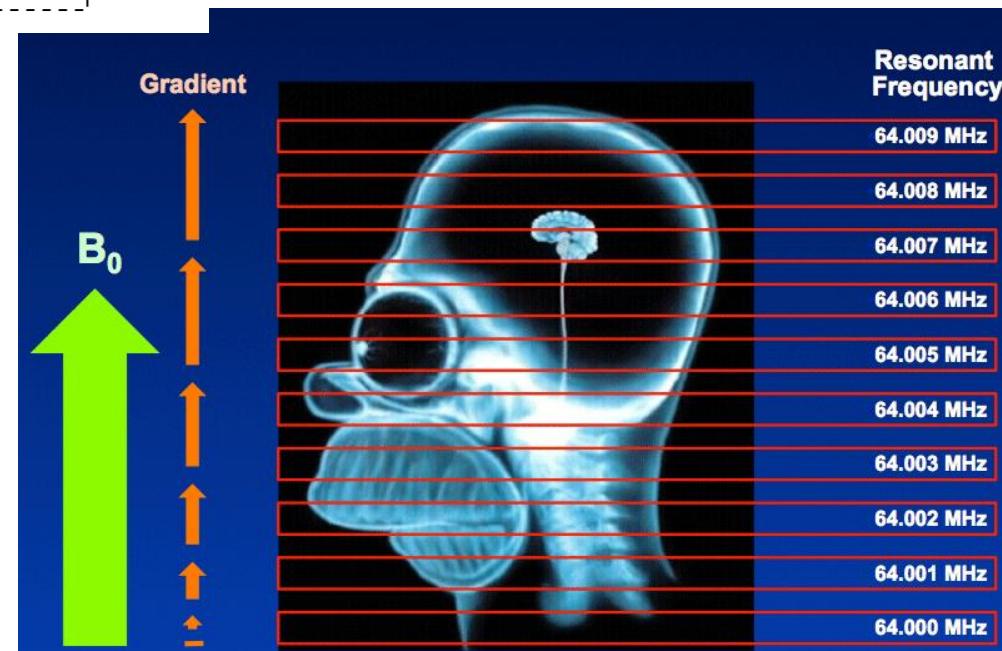
*Gradientní pulsy*

Jiné  $B_0$  - jiná Larmorova frekvence

*Kódování pozice – frekvence*

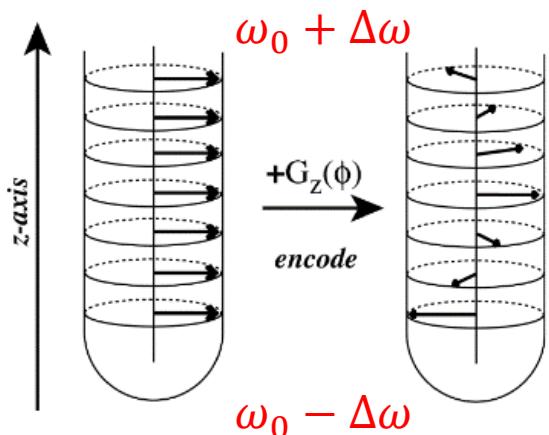
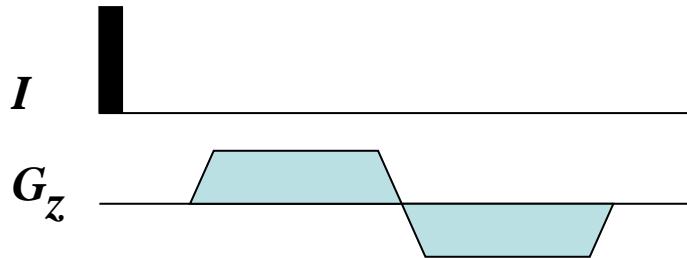
Intenzita signálu na dané frekvenci odpovídá počtu jader na dané pozici

*ZOBRAZOVÁNÍ*



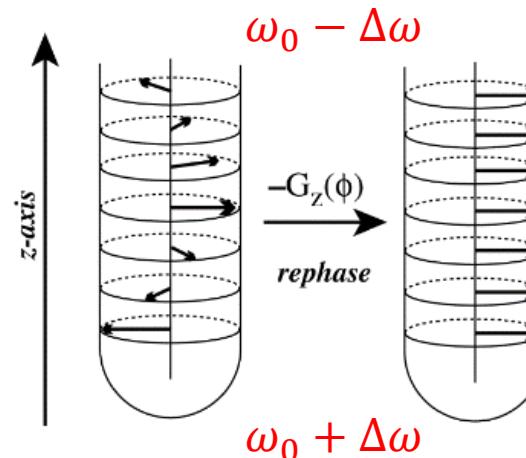
# Využití gradientů v NMR

Gradientní echo



*dephasing*

*Změna  
polarity  
gradientu*



*re-phasing*

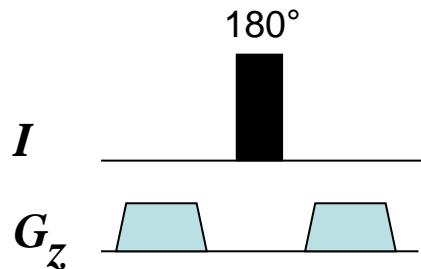
NMR signál (koherence)  
efektivně potlačen

NMR signál (koherence)  
opět viditelný

*ECHO*

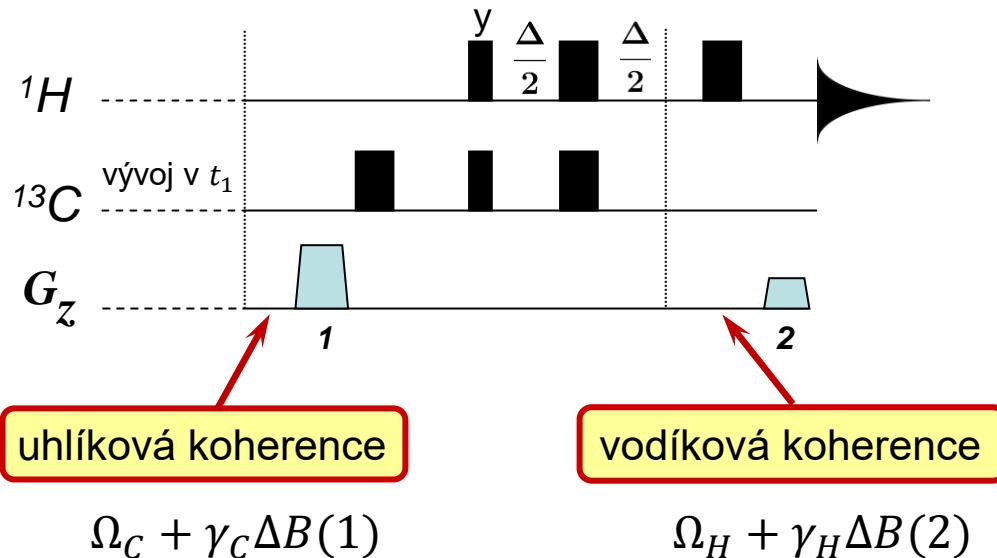
# Využití gradientů v NMR

Potlačení nedokonalostí pulsu



- Změnu polarity zajišťuje  $180^\circ$  puls
- Při nedokonalém pulsu nedojde ke kompletní inverzi
- Gradienty zajistí, že se refokusuje jen invertovaná složka, ostatní se nesfázují a jsou tak potlačeny

Kontrola přenosu koherencí v HSQC experimentu



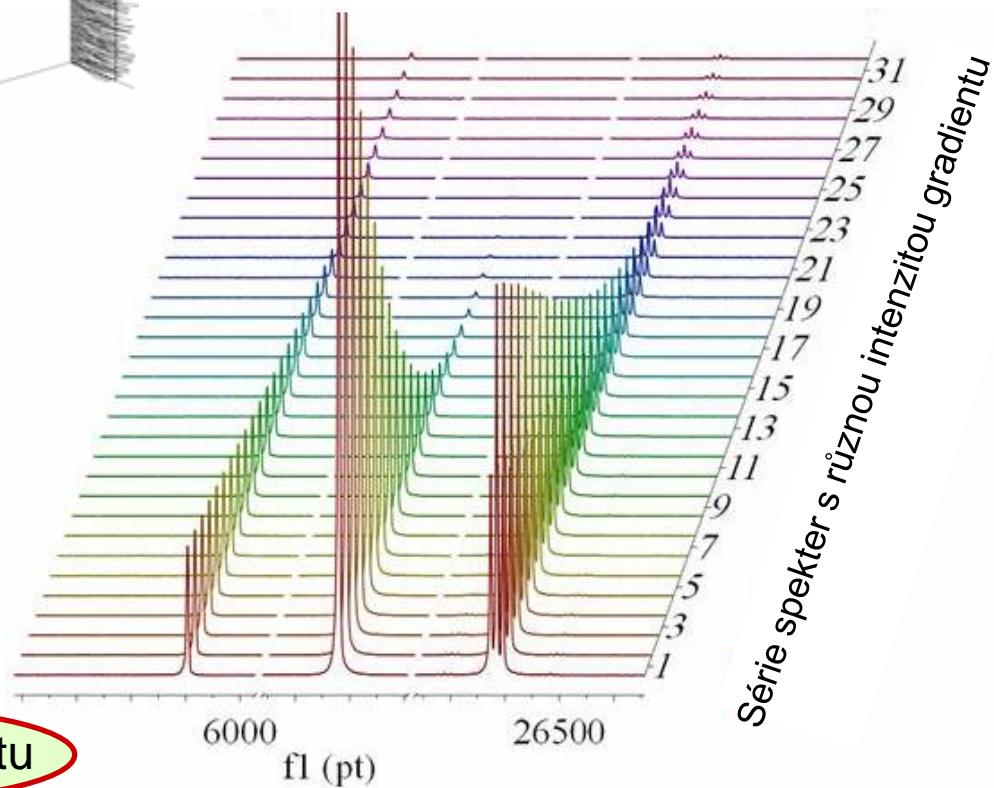
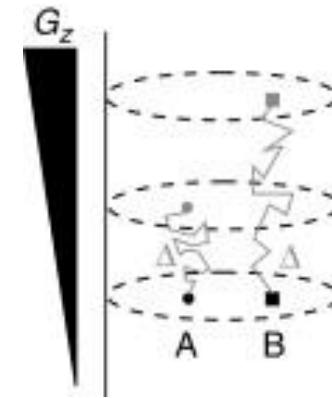
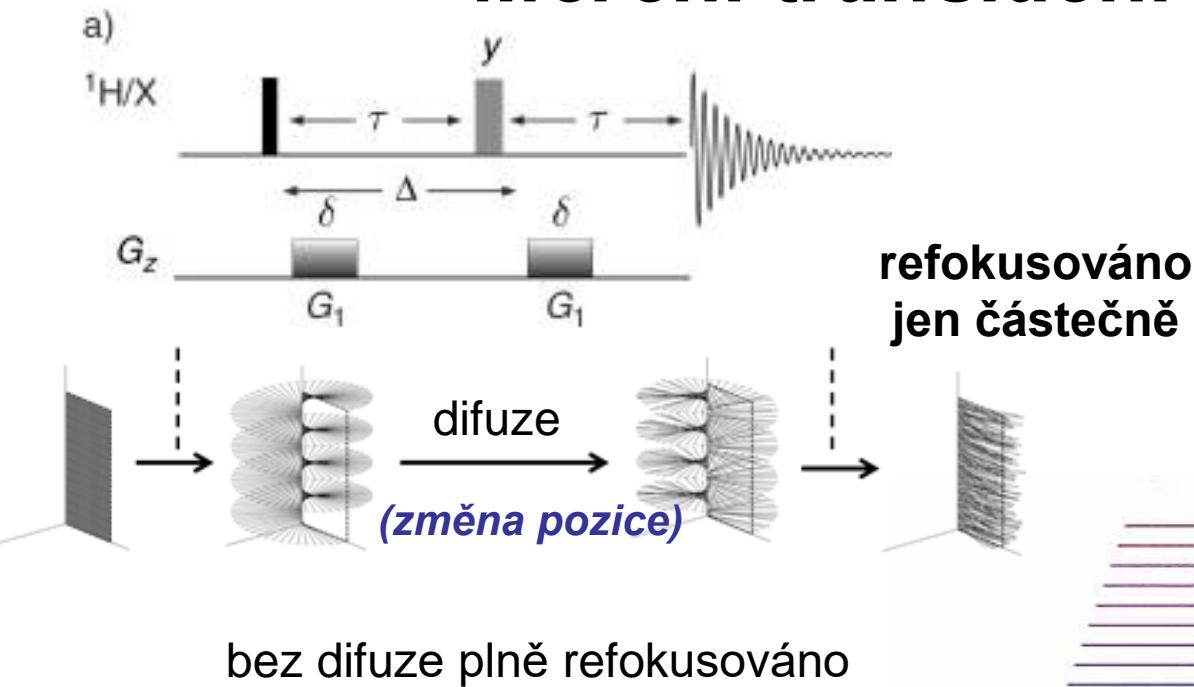
Pro účinný přenos musí platit

$$\frac{G_{z1}\tau_1}{G_{z2}\tau_2} = \frac{\gamma_H}{\gamma_C}$$

Vše ostatní je potlačeno

*De-/re-phasing ovlivněn  
gyromagnetickým poměrem*

# Měření translační difuze



Stejskal-Tannor

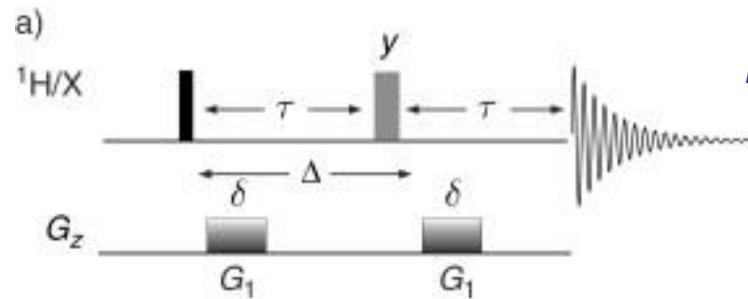
$$I(g) = I_0 \exp \left\{ -D\gamma^2 g^2 \delta^2 \left( \Delta - \frac{\delta}{3} \right) \right\}$$

difúzní  
koeficient

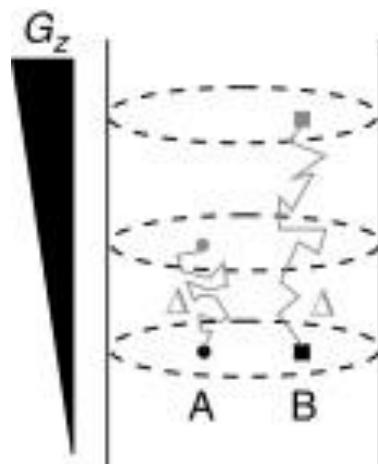
Intenzita gradientu

# Translační difuze - DOSY

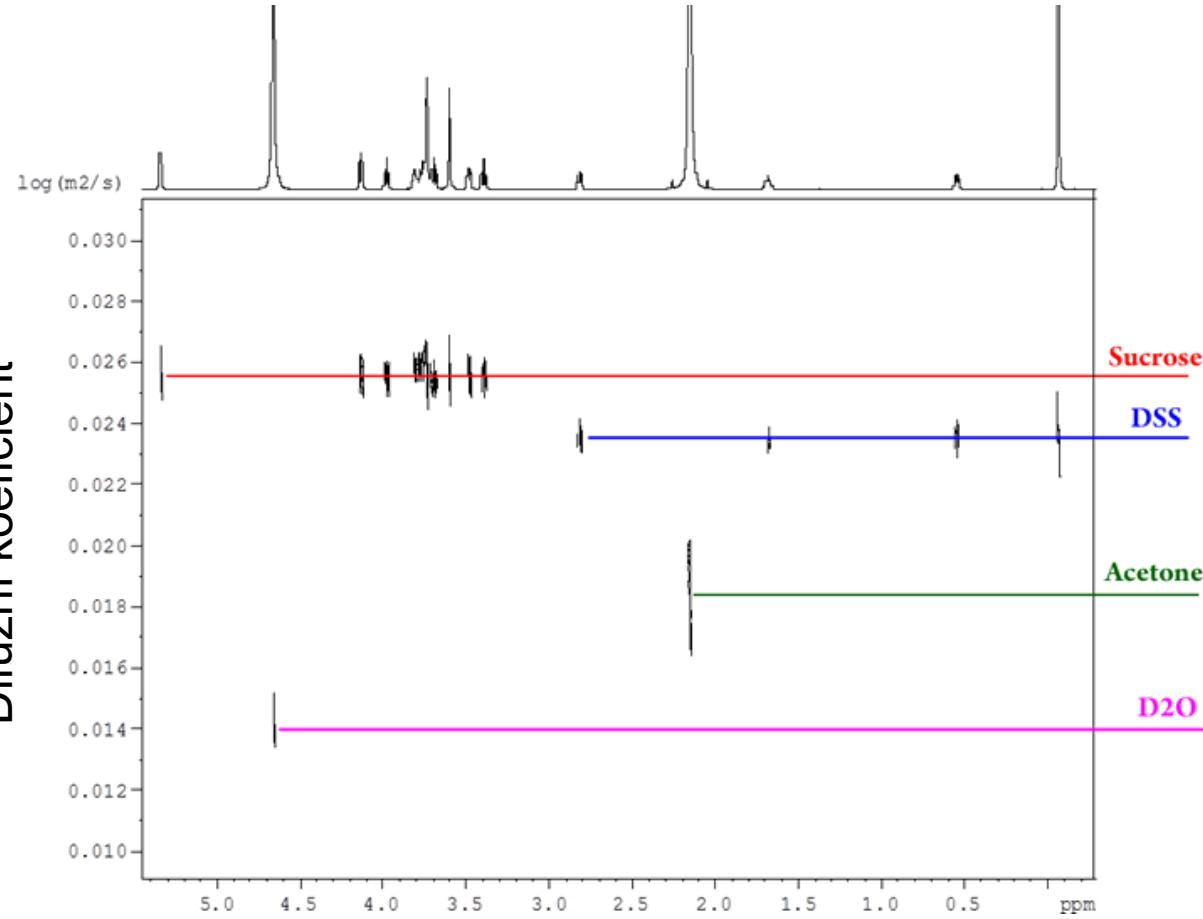
Diffusion Ordered SpectroscopY



*Lze rozlišit jednotlivé složky směsi dle velikosti molekul*



Difúzní koeficient



# Translační difuze - informace

$$D = \frac{k_B T}{f}$$

viskozita

$$f = 6\pi\eta r_S$$

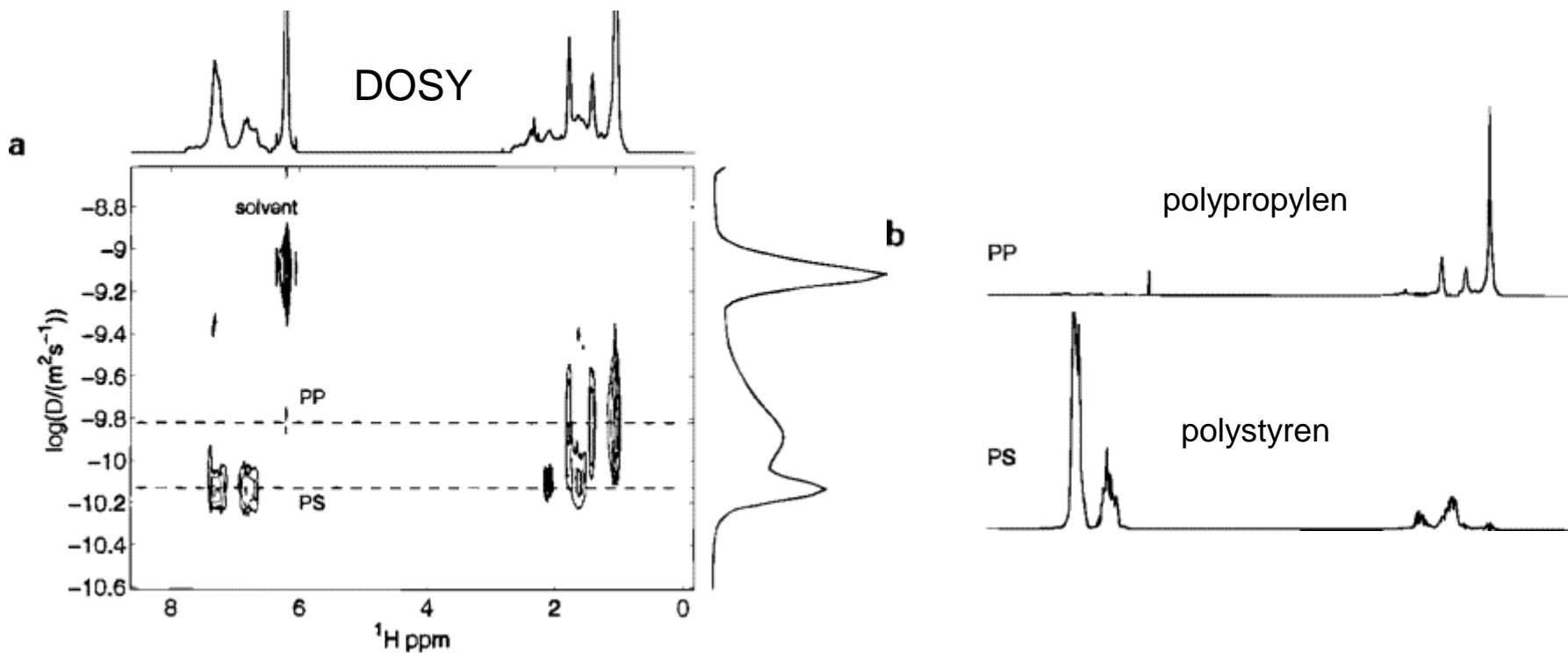
$$V = \frac{4\pi}{3} r^3$$

hydrodynamický poloměr, objem

$$D \propto \frac{1}{\sqrt[3]{M}}$$

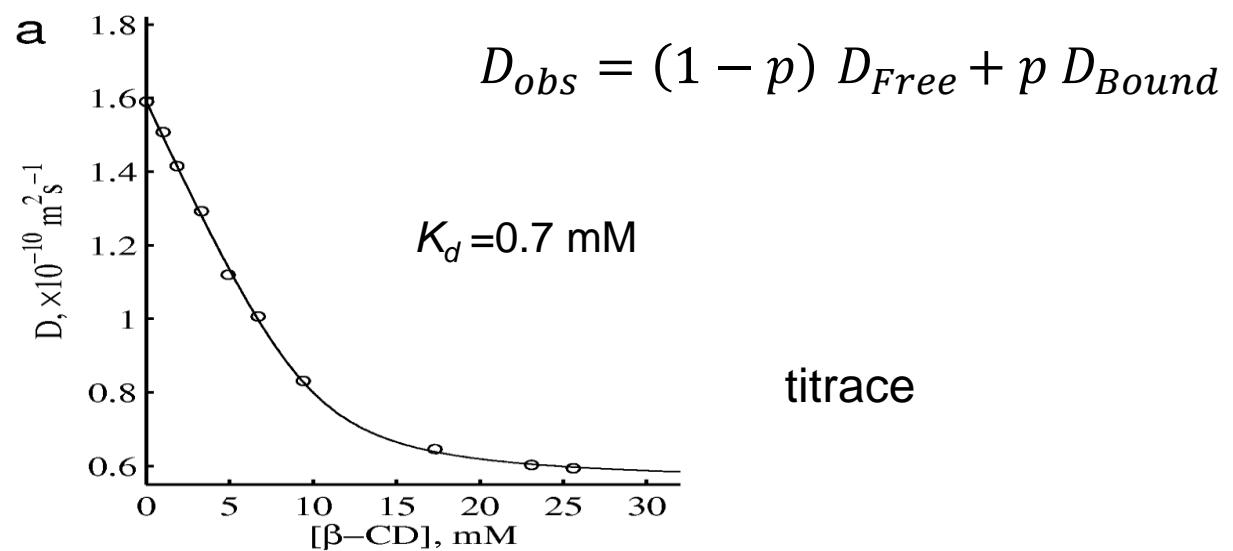
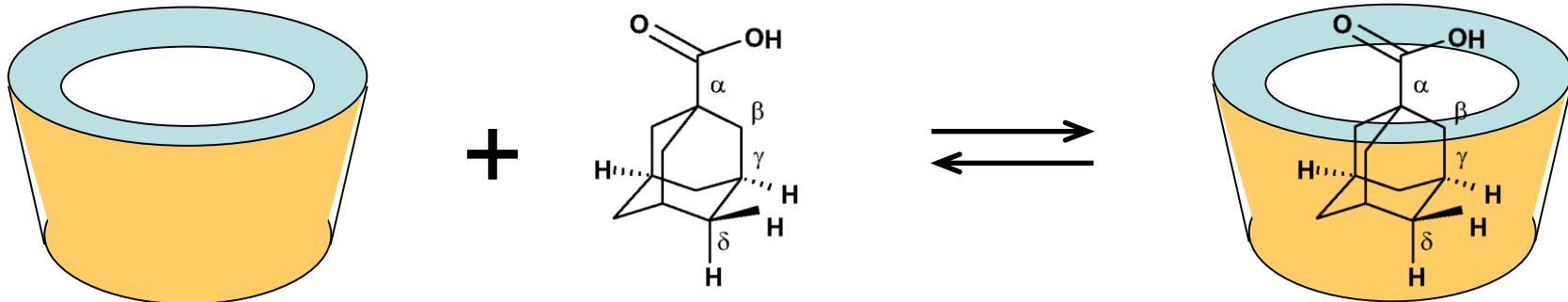
molekulová hmotnost

Komplexy, agregace, tvorba micell, solvatace, polydispersita polymerů



# Translační difuze a výměna

Tvorba komplexu



# **NMR of biomolecules**

- highly selective response, can resolve individual atoms
- broad range of physicochemical characteristics
- molecules in solution – closer to physiological state
- insoluble structures, amyloids, membrane proteins – solid-state NMR

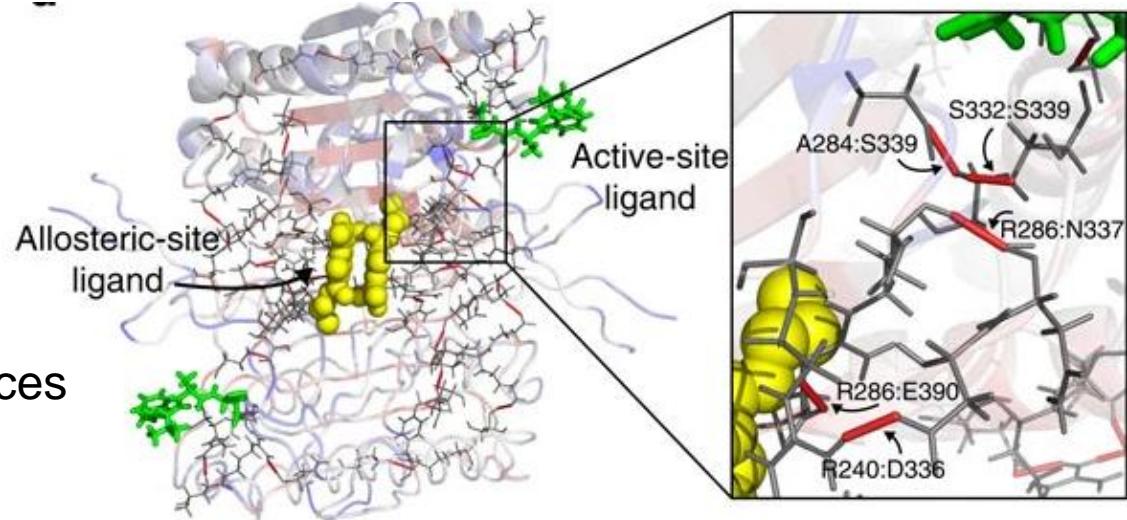
*peptides and proteins*

*nucleic acids*

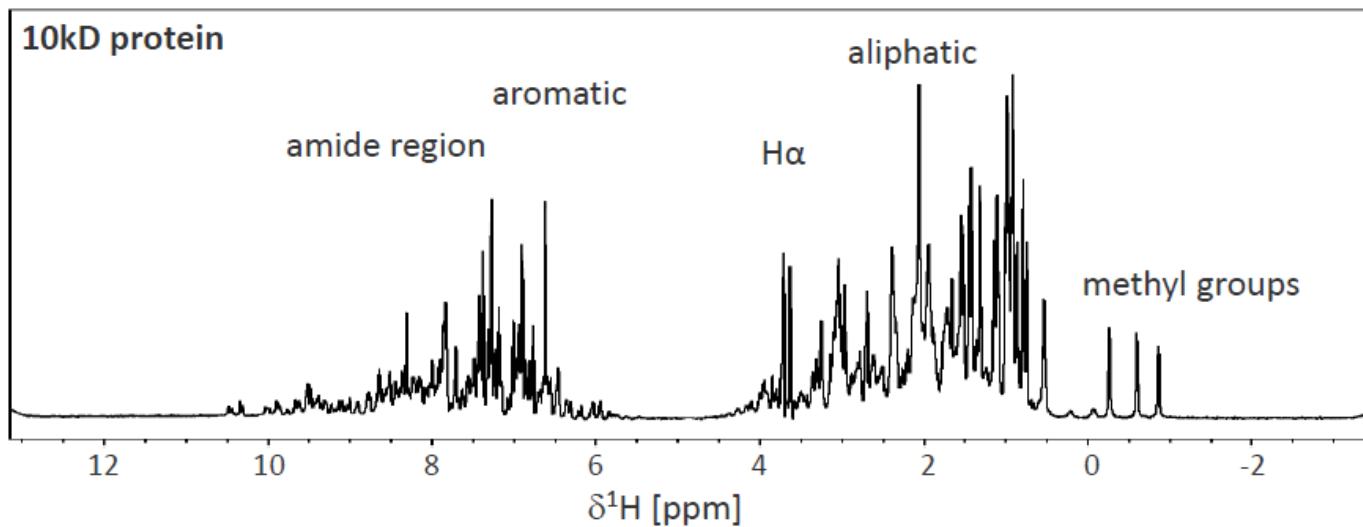
*oligosaccharides*

## Available information

- 3D structure
- dynamics
- substrate identification
- complexation
- interaction surfaces / interfaces
- chemistry of active site
- protein folding process
- ...



# Protein NMR



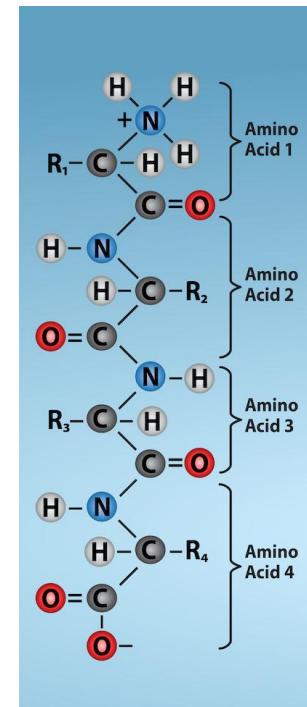
atoms in proteins	<u>stable</u> Isotope composition (natural abundance)
Hydrogen (H)	$^1\text{H}$ (99.9885%) $^2\text{H}$ (0.0115%)
Carbon (C)	$^{12}\text{C}$ (98.93%) $^{13}\text{C}$ (1.07%)
Nitrogen (N)	$^{14}\text{N}$ (99.632%) $^{15}\text{N}$ (0.368%)

Spectral crowding resolved in multidimensional spectra

$^{13}\text{C}/^{15}\text{N}$  ( $^{13}\text{C}/^{15}\text{N}/^2\text{H}$ ) labelled samples

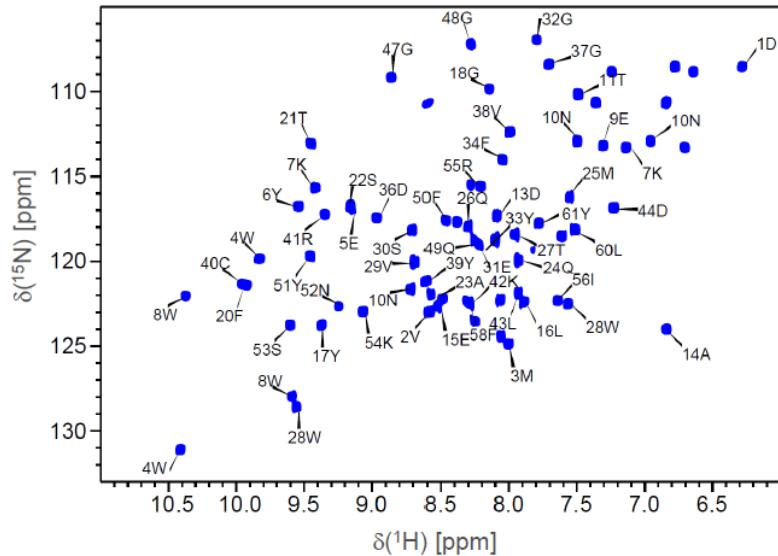
Protein expression in bacteria cells

- minimal media ( $^{15}\text{NH}_4\text{Cl}$ ,  $^{15}\text{NH}_4\text{SO}_4$  – solely nitrogen source,  $^{13}\text{C}$ -glukose,  $^{13}\text{C}$ -glycerol – solely carbon source)

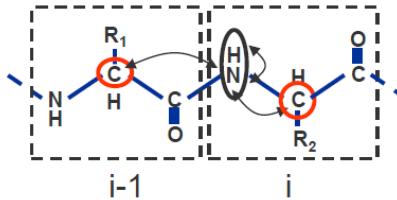


# Multidimensional approaches

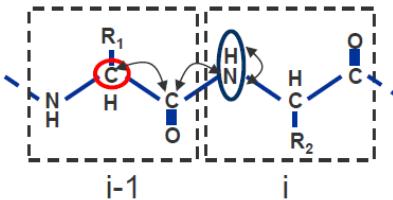
$^1\text{H}$ - $^{15}\text{N}$  2D HSQC spectrum = protein fingerprint



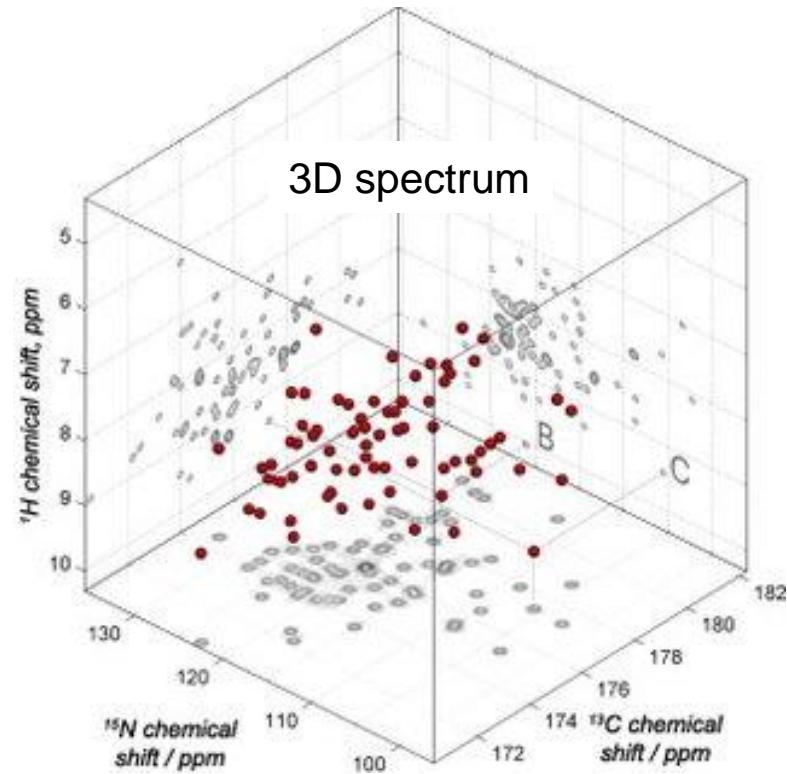
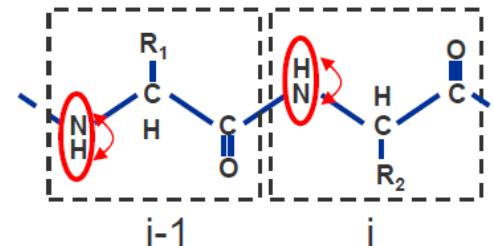
**HNCA**  
signal for  $\text{C}\alpha$  of  $\text{NH}(i)$  and  $\text{NH}(i-1)$



**HN(CO)CA**  
signal for  $\text{C}\alpha$  of  $\text{NH}(i-1)$



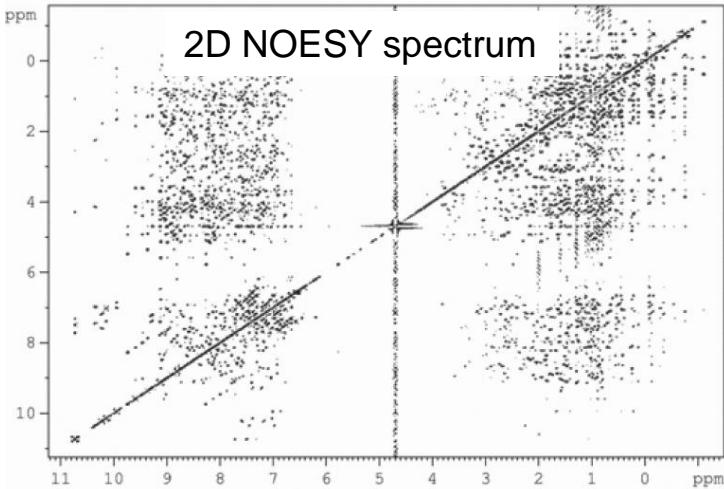
*Each amino acid represented with one peak*



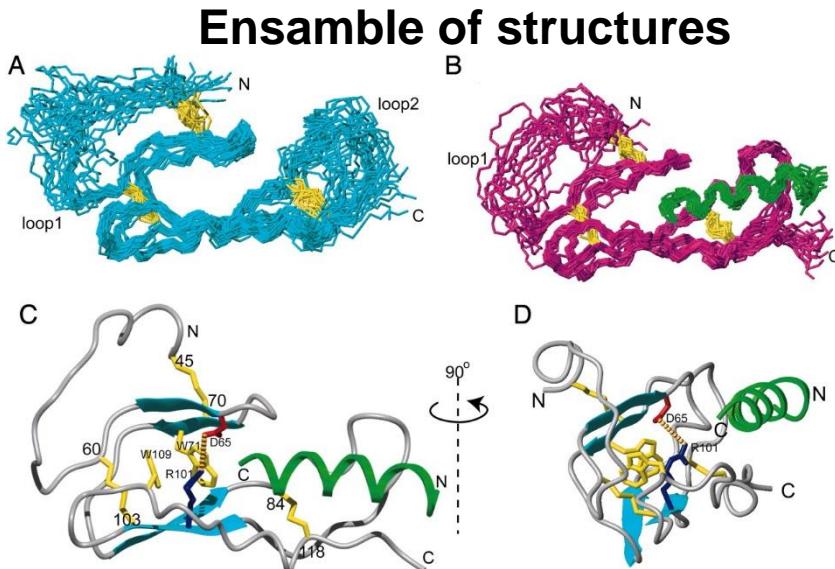
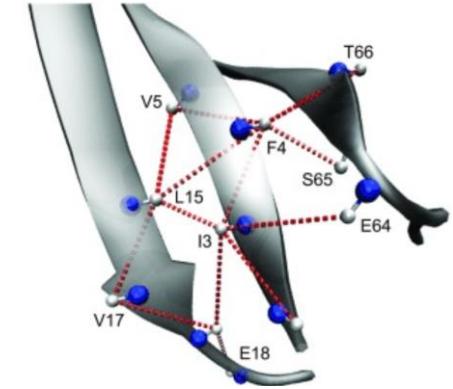
# Structural information

Interpreting NOESY crosspeaks

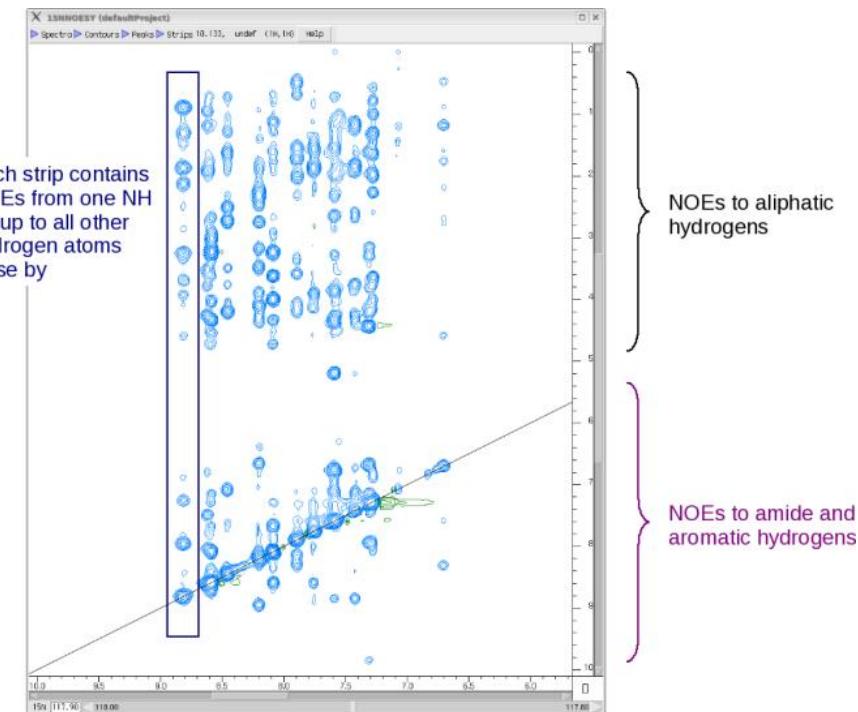
Crosspeak volume  $\longleftrightarrow$  distance



- Thousands of restraints
- Minimization procedure



3D NOESY-HSQC spectrum



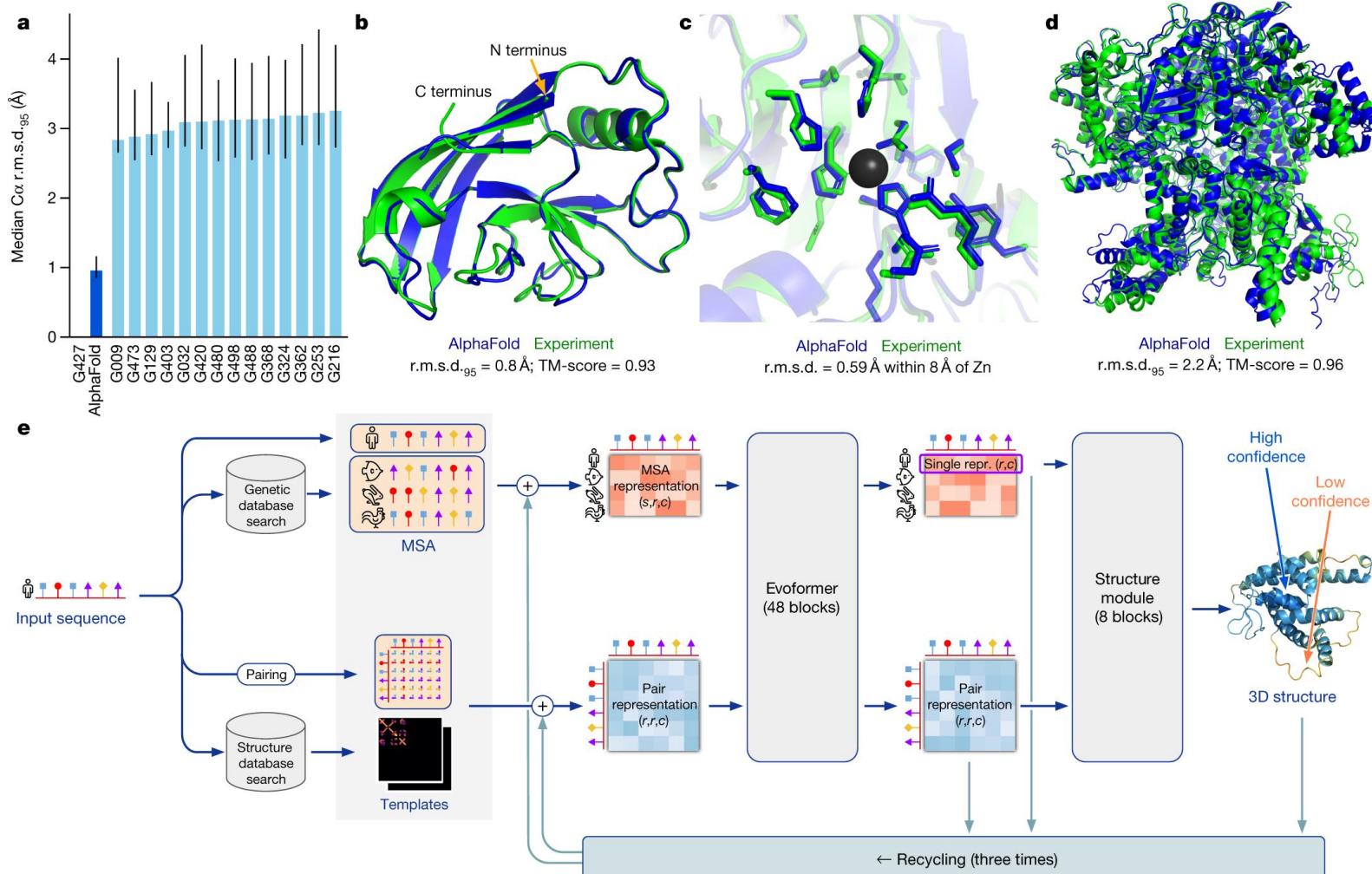
# AlphaFold

## Protein Structure Database

Developed by Google DeepMind and EMBL-EBI

Search for protein, gene, UniProt accession or organism or sequence search BETA Search

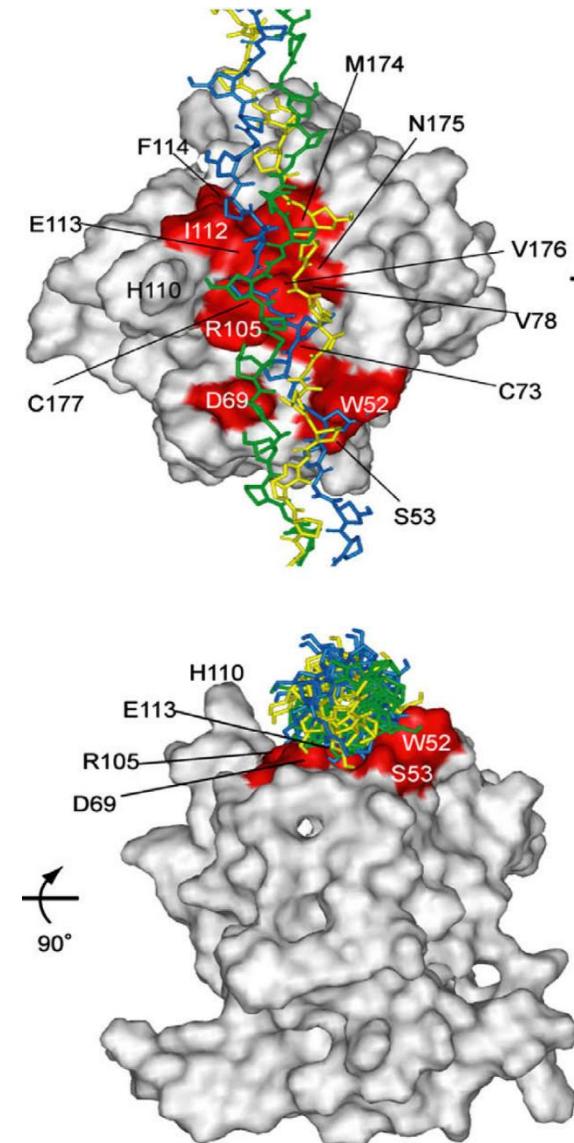
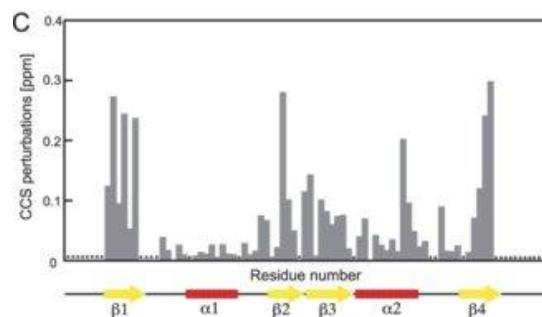
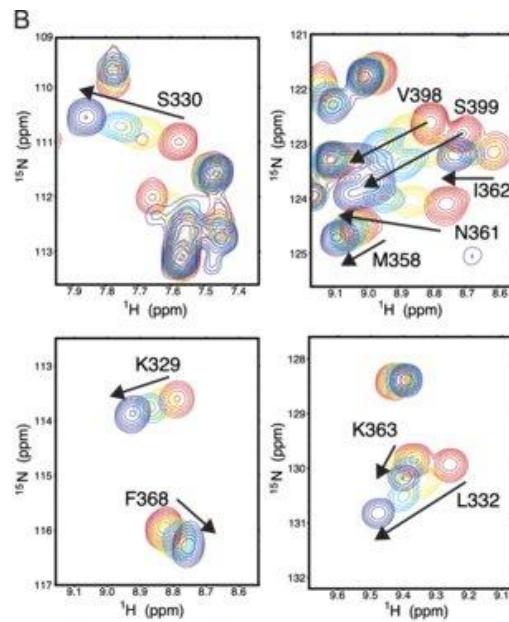
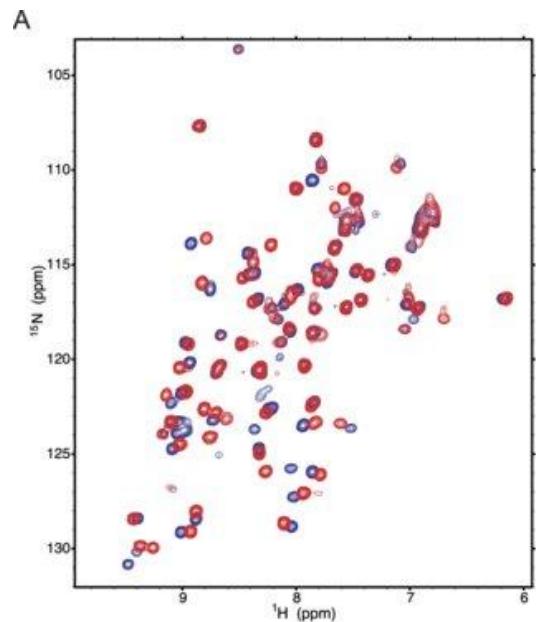
Examples: MENFQKVEKIGEGTYGV... Free fatty acid receptor 2 At1g58602 Q5VSL9 E. coli



# *Monitoring protein interactions*

Chemical shift mapping during titration

$^1\text{H}$ - $^{15}\text{N}$  HSQC



Fast Exchange between free and bound states averages chemical shifts

# Monitoring protein interactions

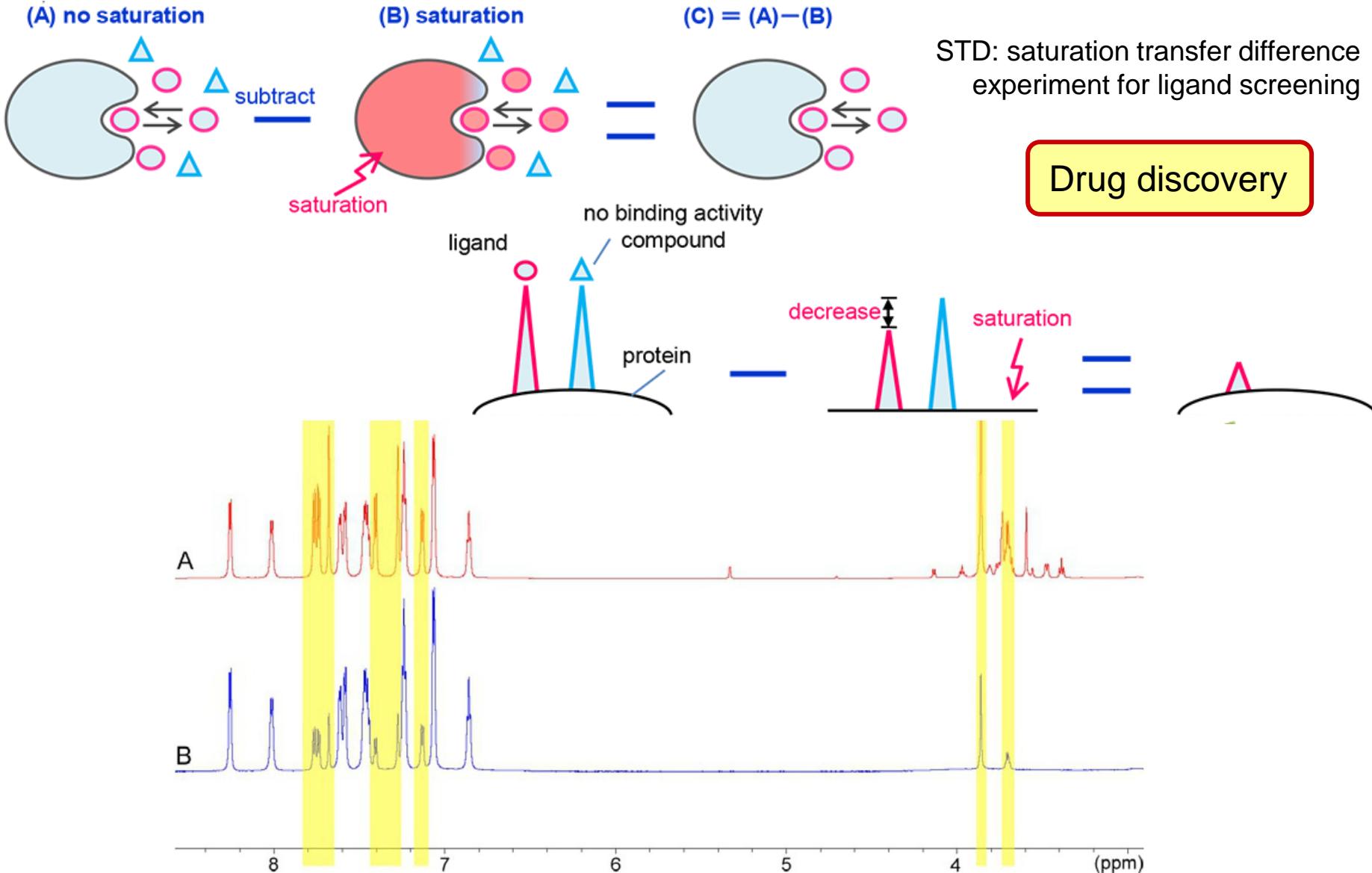


Fig. 7 (A) Reference spectrum of a sample containing 22  $\mu\text{M}$  HSA and a cocktail of (6 mM each): Naproxen, ANS, and sucrose. Peaks highlighted in yellow belong to Naproxen. Remaining peaks above 6 ppm belong to ANS, and remaining peaks below 6 ppm

belong to sucrose. (B) STD NMR difference spectrum of the same sample, showing the presence of Naproxen and ANS signals. This shows that Naproxen and ANS are ligands for HSA, whereas sucrose is not.

# ***Introduction to solid-state NMR***

## **Samples**

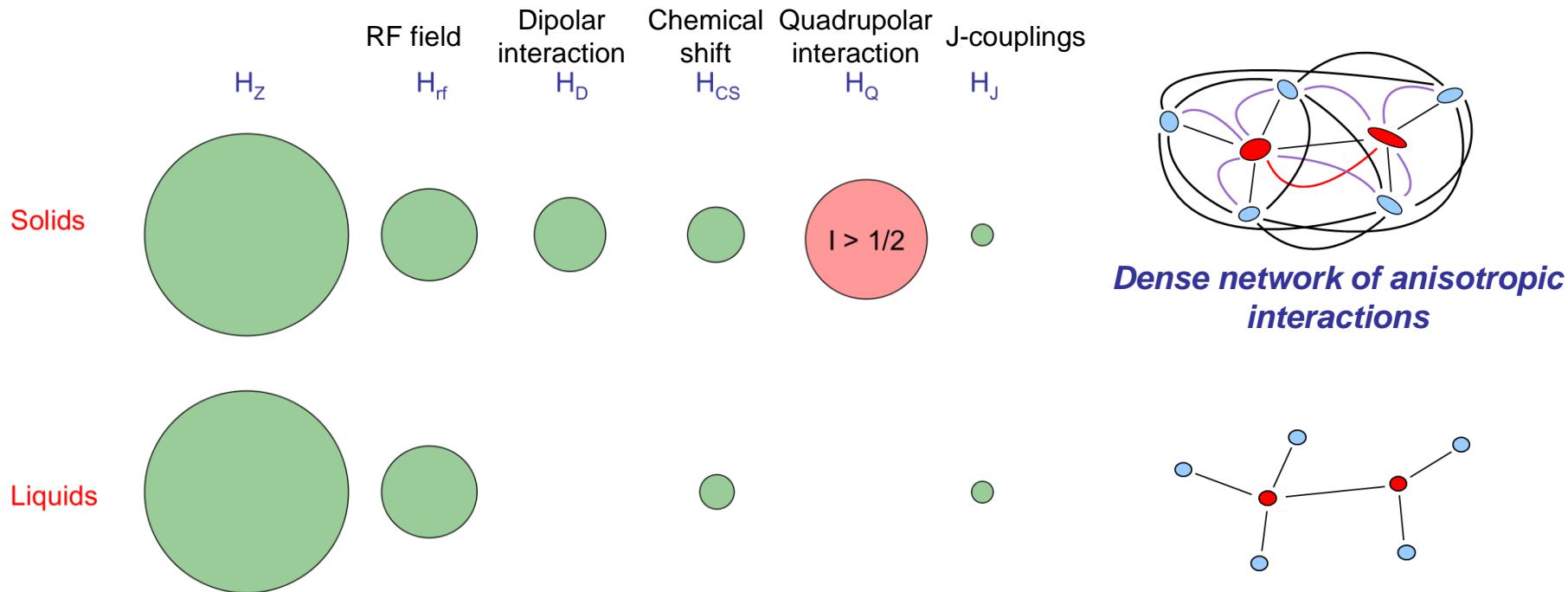
crystalline solids, polymers, amorphous materials, glasses, functional and porous materials (zeolites), insoluble proteins, large molecular assemblies,....

## **Methods**

Magic angle spinning, cross-polarization, decoupling and recoupling methods, multiple-quantum correlations, wide-line excitations, magnetization transfers, multidimensional ...

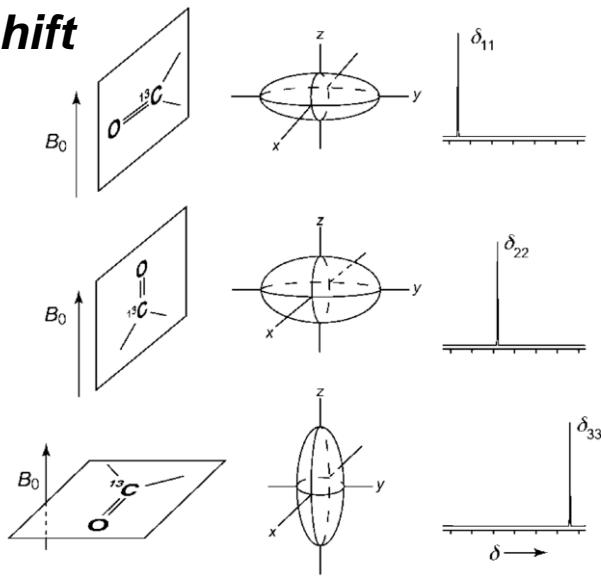
## **Information**

Molecular conformation, non-covalent interactions, oxidation state or chemical reactivity, morphology of porous solids, properties of catalytic centers, dynamics in confinement, NMR crystallography, interpretation of electronic environment using DFT calculations, ...

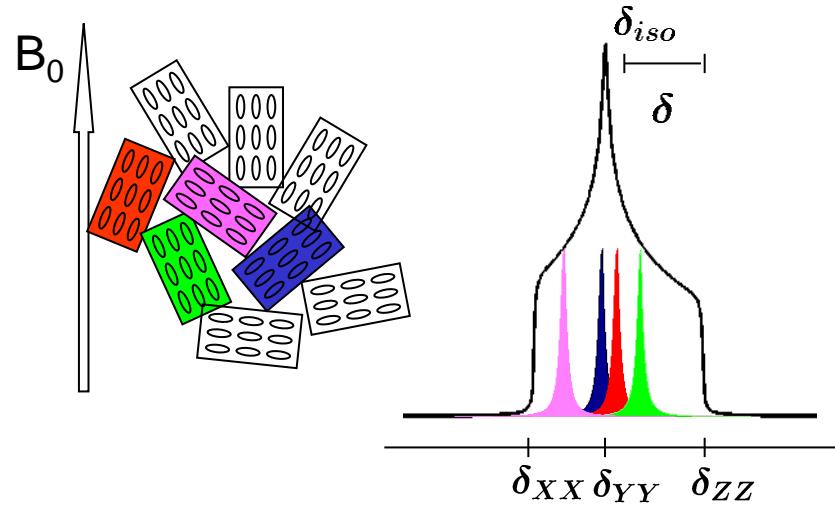


# Anisotropy of interactions in solids

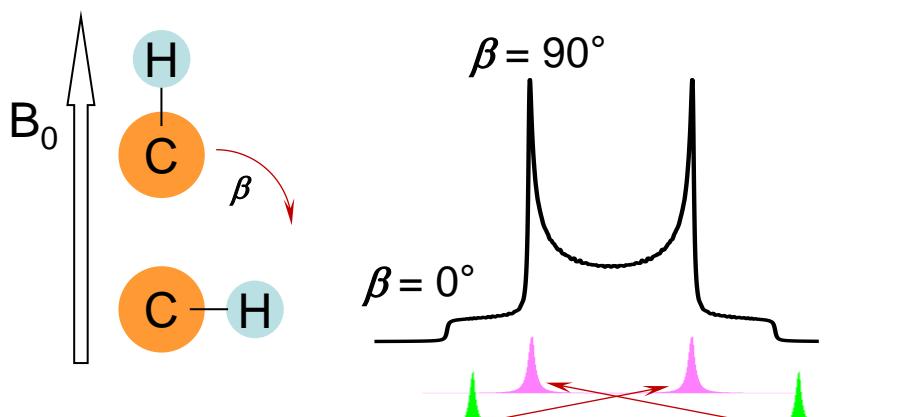
## Chemical shift



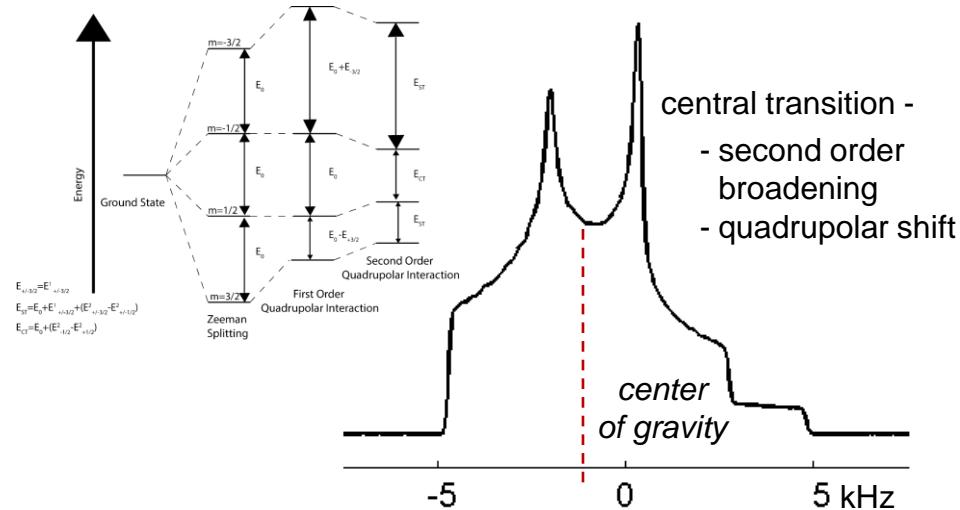
## Powder sample



## Dipolar interactions

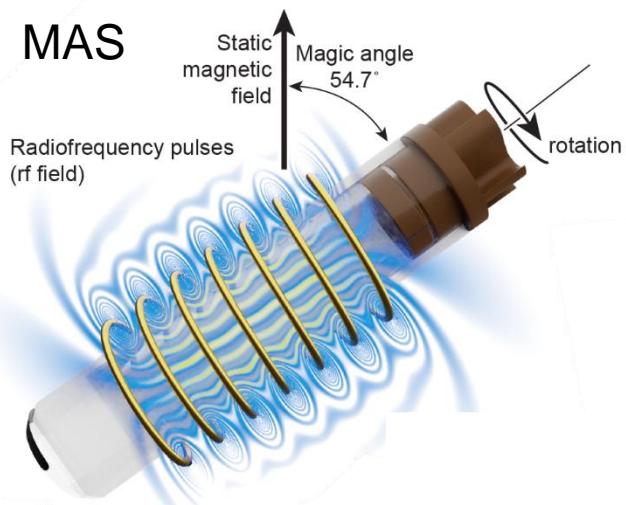


## Quadrupolar interactions

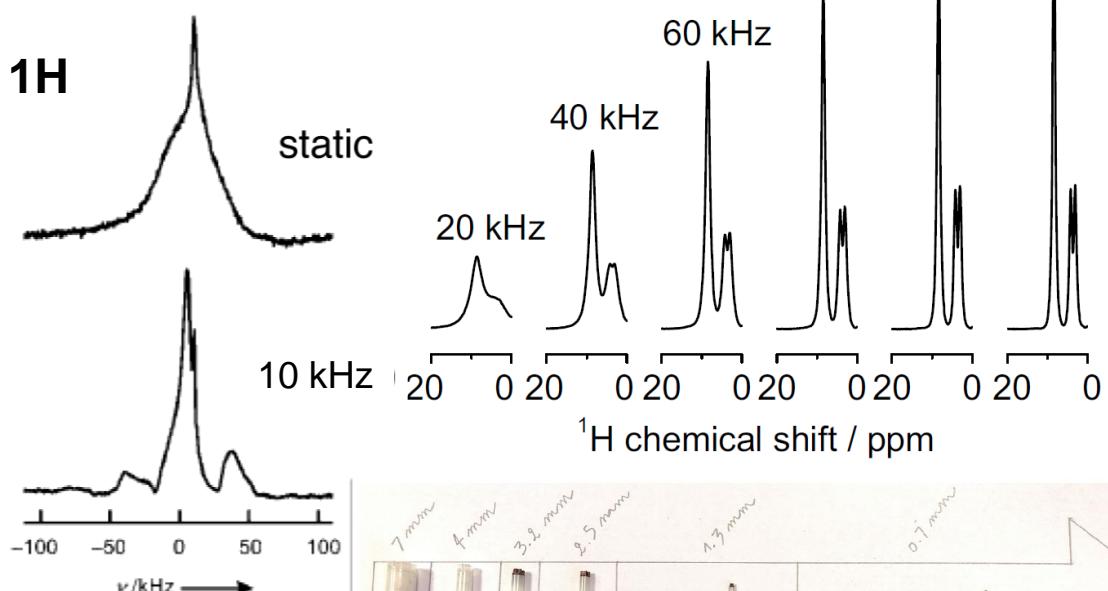


# Recovering resolution – Magic Angle Spinning

MAS



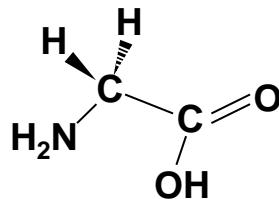
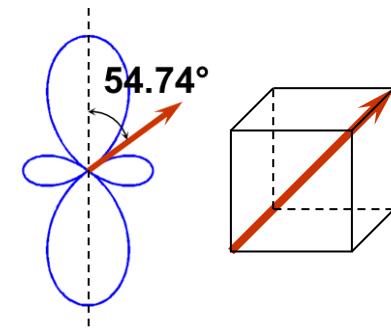
$^1\text{H}$



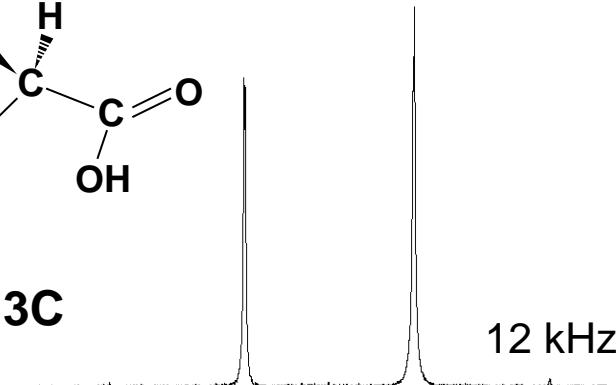
$$H_{int} = \begin{matrix} \text{spatial} \\ \text{dependence} \end{matrix} \cdot \begin{matrix} \text{spin} \\ \text{operators} \end{matrix}$$

*averaged to*

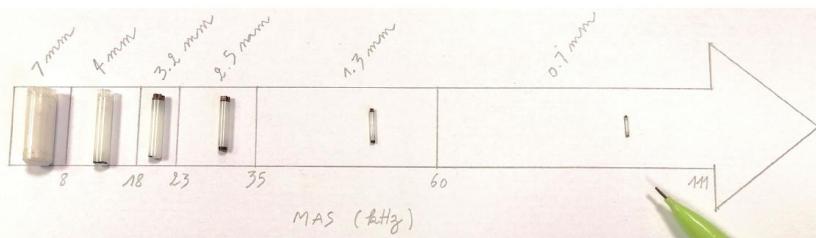
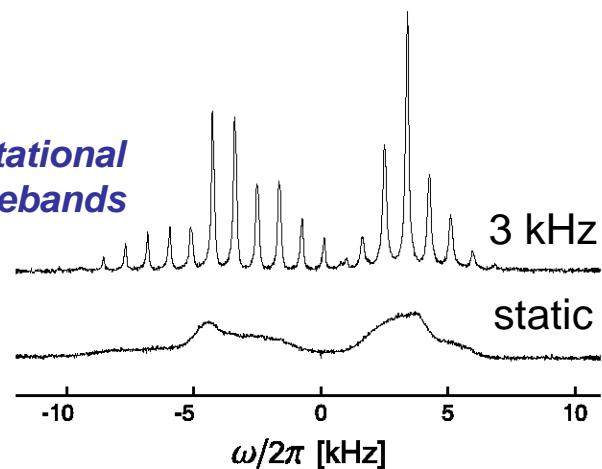
$$\frac{1}{2}(3 \cos^2 \beta_M - 1)$$



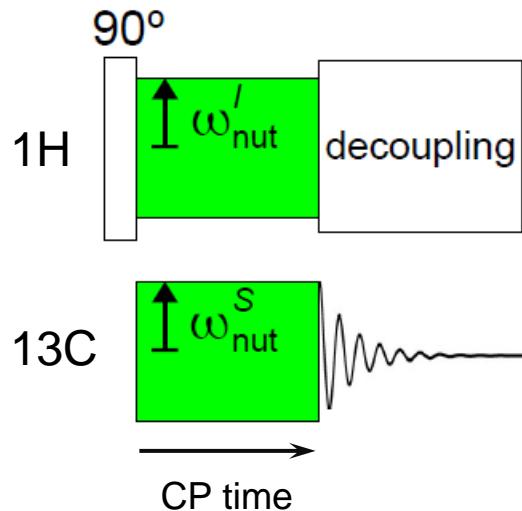
$^{13}\text{C}$



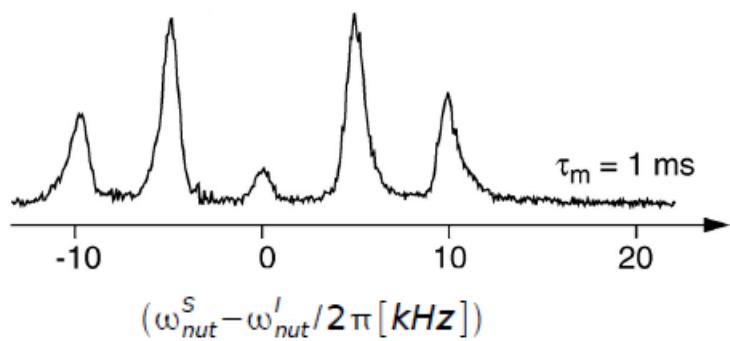
*Rotational sidebands*



# Enhancing signal – Cross-polarization

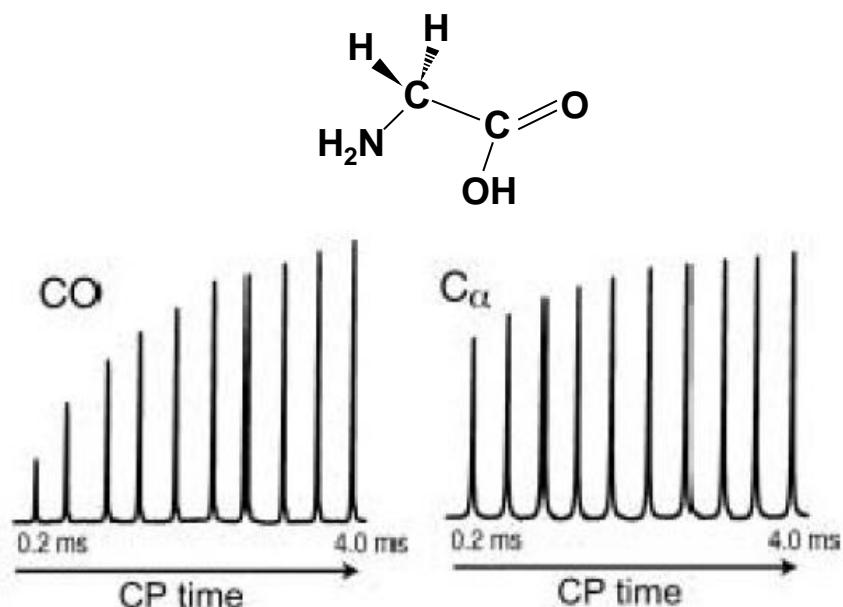
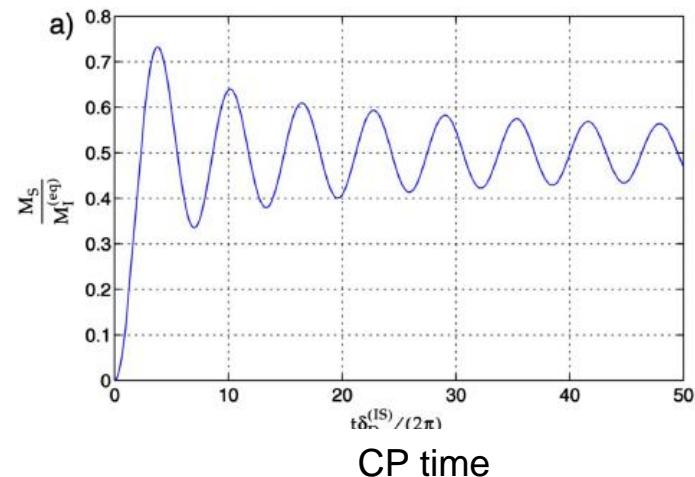


Polarization transfer from  
1H to 13C mediated by  
direct dipole-dipole  
interaction



Hartmann-Hahn  
match

$$\omega_{\text{nut}}^I = \omega_{\text{nut}}^S \pm n \omega_r$$



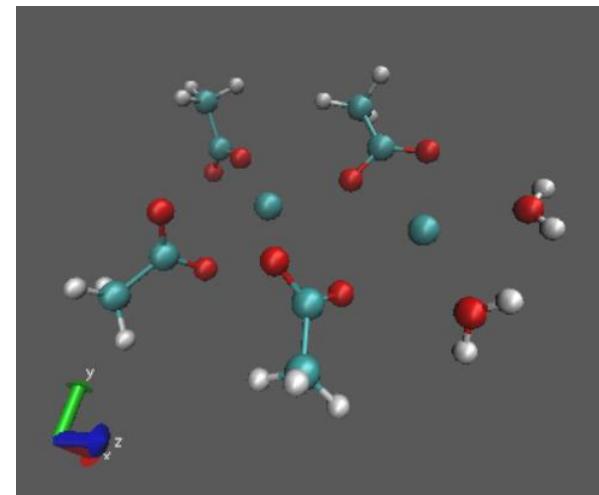
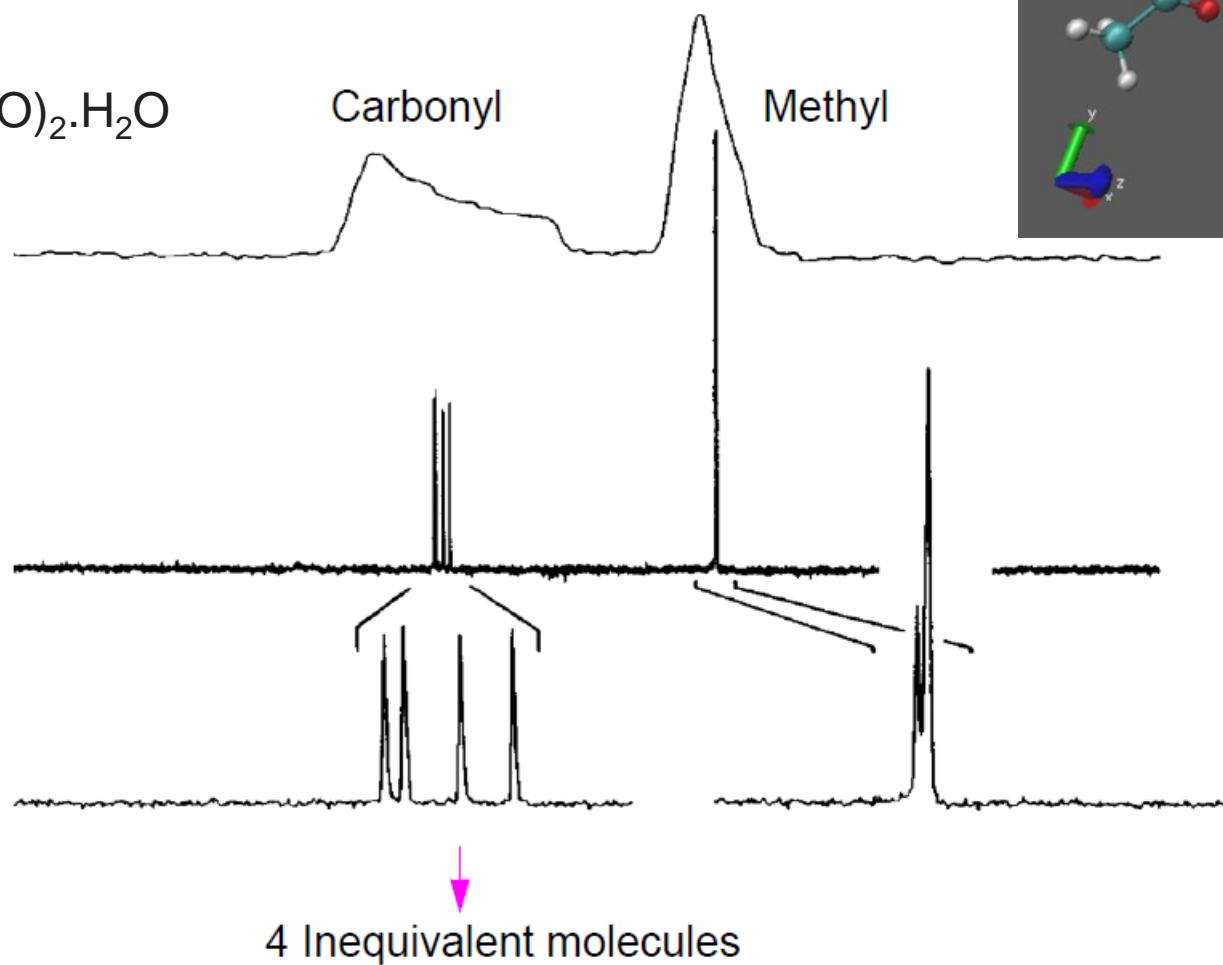
Buildup rate depends on dipolar coupling strength = distance

# *Non-equivalent positions in crystal*



Carbonyl

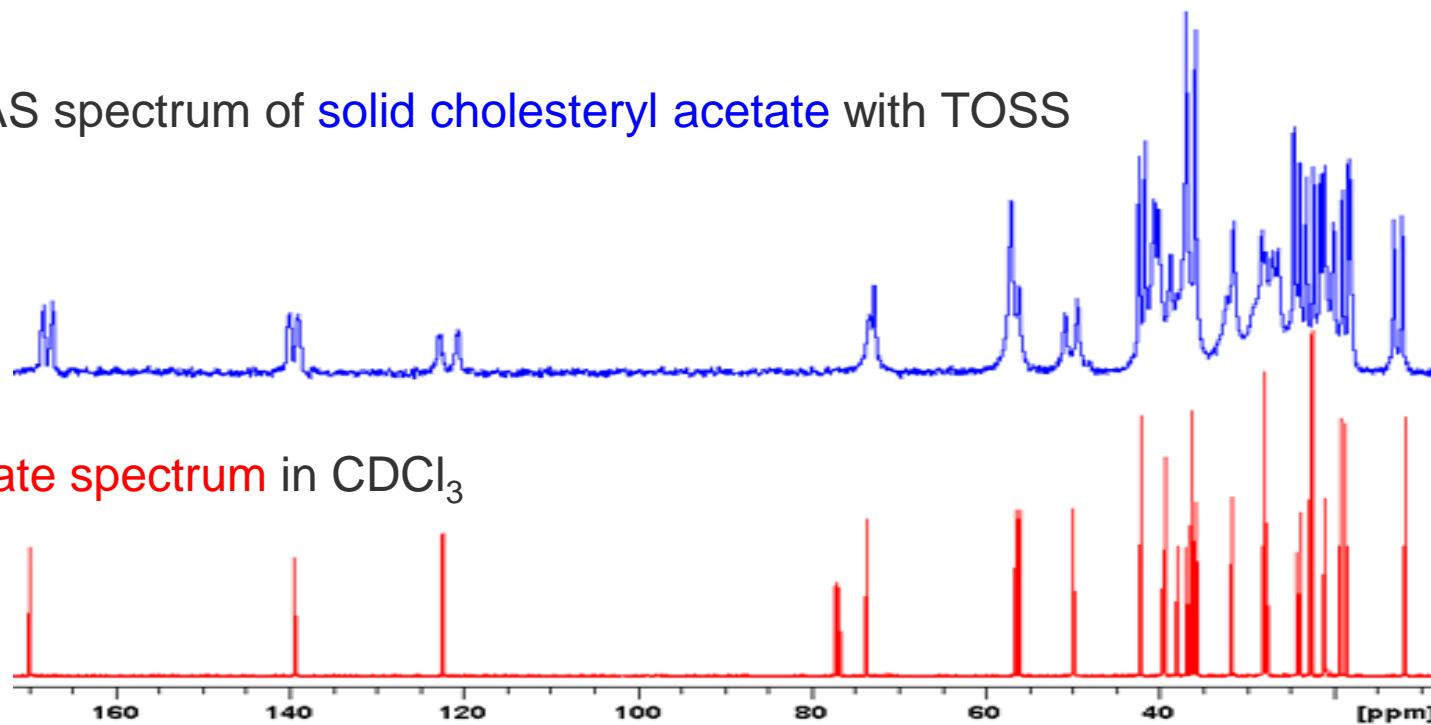
Methyl



# *Přiřazení signálů a INADEQUATE*

- přenesení ze spektra roztoku
- INADEQUATE
- další 2D SQ-DQ techniky (RFDR, C7, ...)

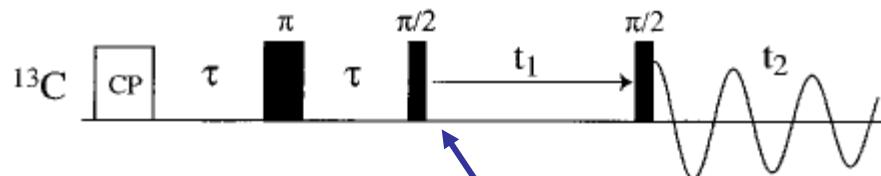
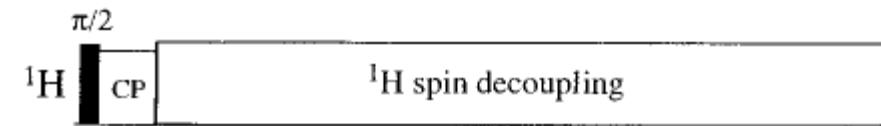
$^{13}\text{C}$  CP-MAS spectrum of solid cholesteryl acetate with TOSS



# Přiřazení signálů a INADEQUATE

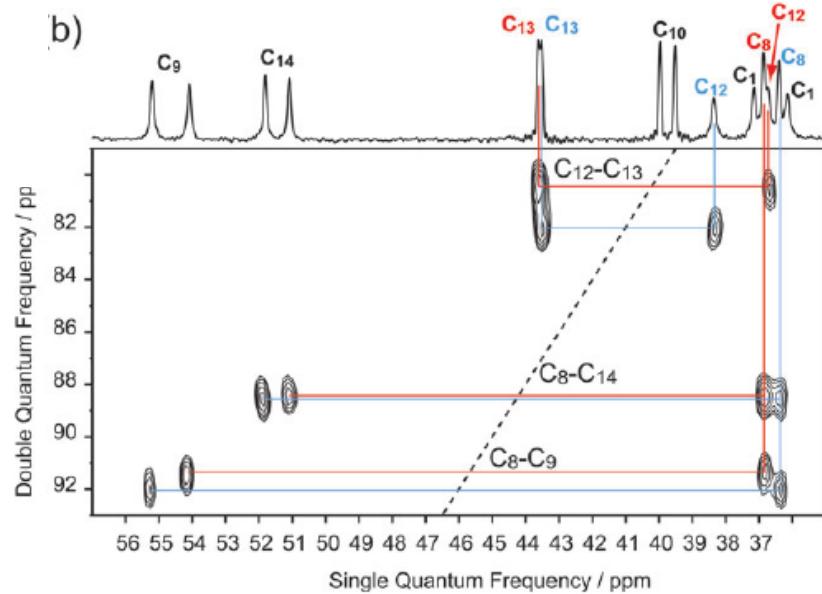
- přenesení ze spektra roztoku
- INADEQUATE
- další 2D SQ-DQ techniky (RFDR, C7, ...)

Incredible Natural Abundance  
Double QUAntum Transfer  
Experiment

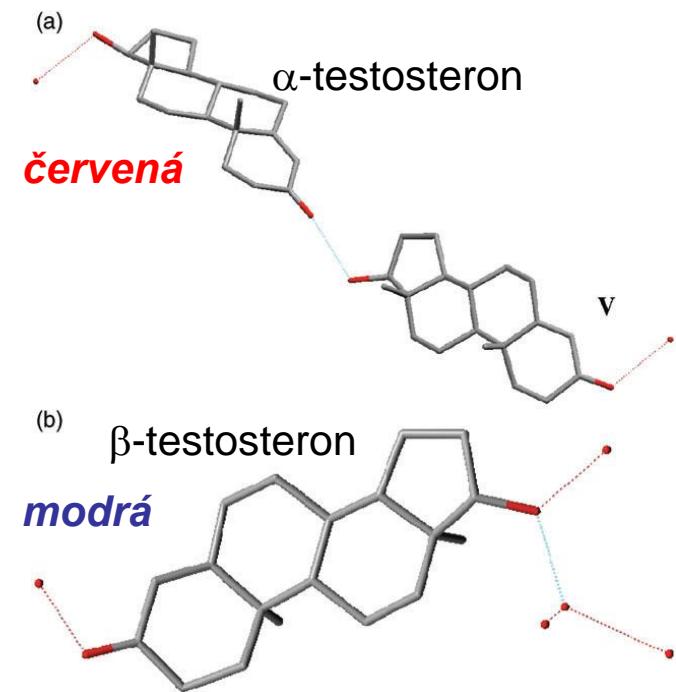
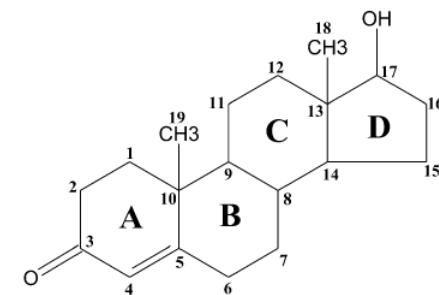
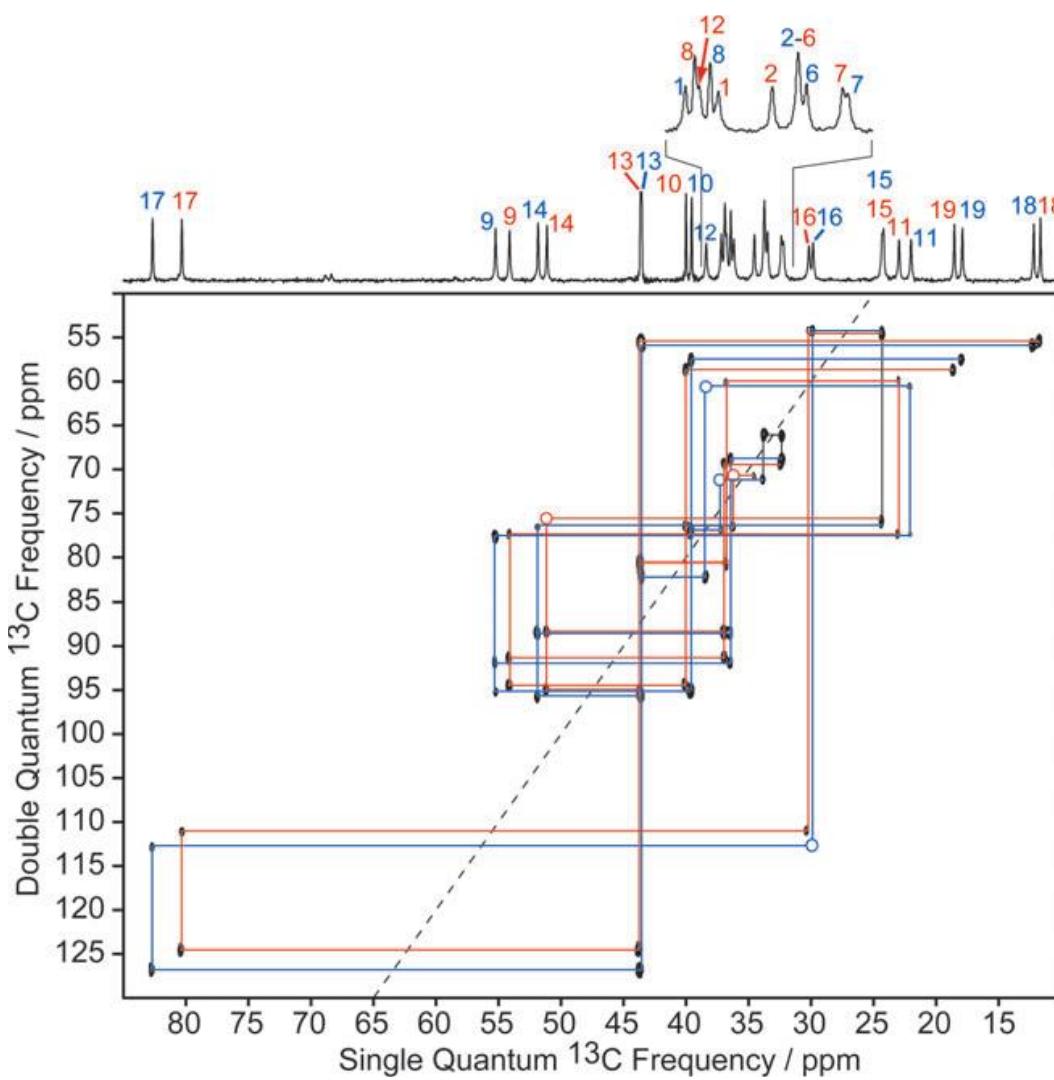


dvoukvantová  
koherence

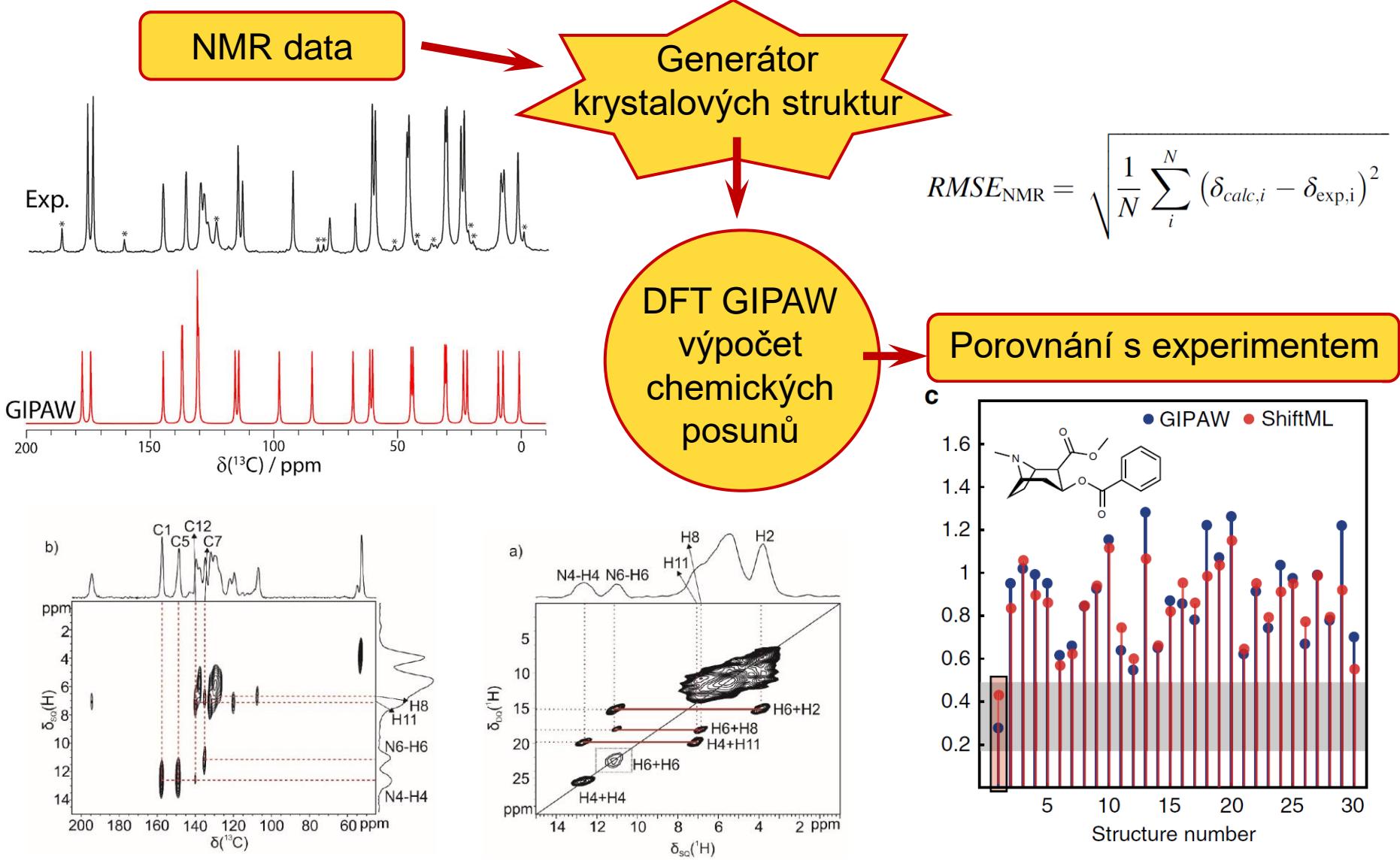
vzniká mezi sousedními uhlíky  
s J vazbou nebo DD interakcí



# *Polymorfismus krystalů*



# NMR krystalografie



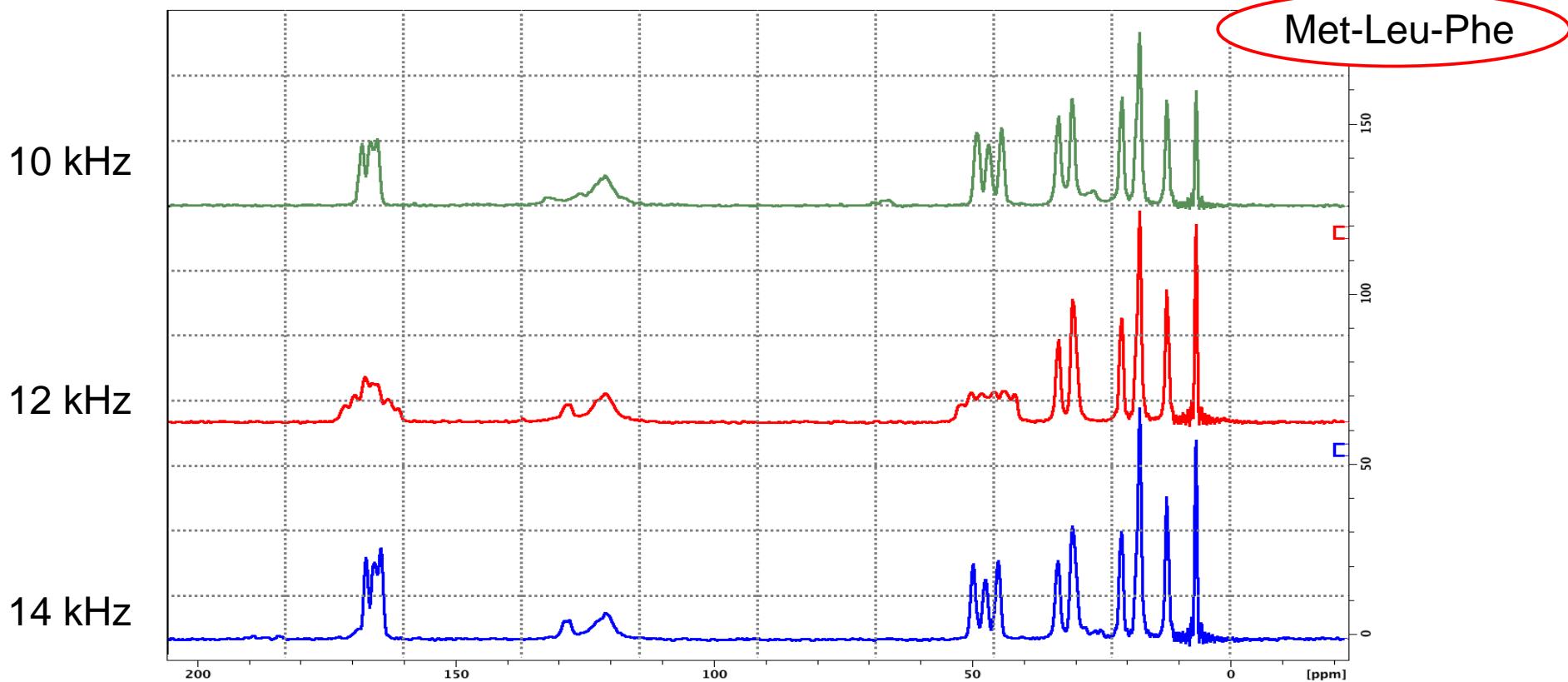
# Homonukleární dipolární recoupling

Rotational Resonance

$$\Delta\omega_{iso} = n \omega_R$$

*spinová rotace  
chem. posun*

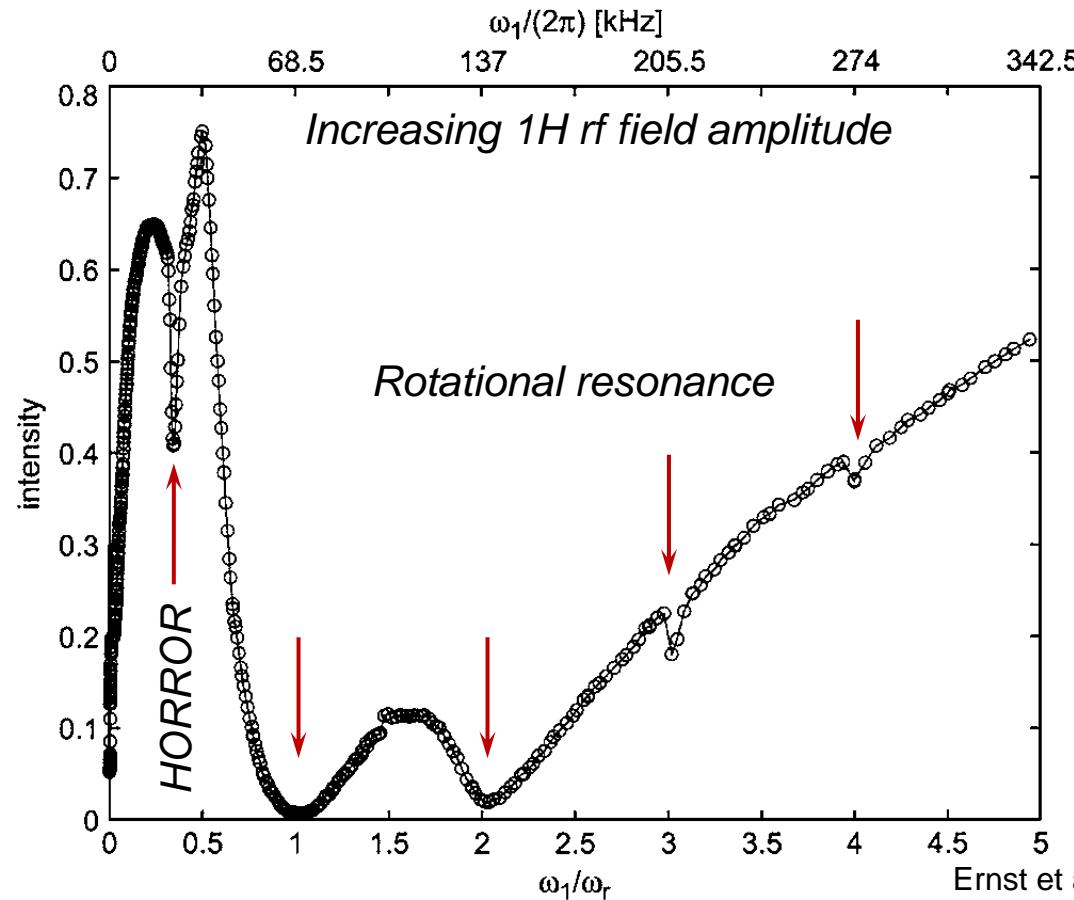
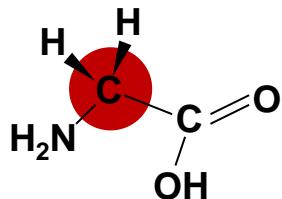
*prostorová rotace  
MAS*



# Heteronuclear decoupling at elevated MAS

$^1\text{H}$  decoupling and resonance conditions

MAS frequency of 68.4 kHz  
1.3 mm rotor  
600 MHz spectrometer



Ernst et al., JCP, 2005, 064102

# Průměrný Hamiltonián a pulzní sekvence

Laboratorní systém

$$e^{-iH_Z t}$$

$$H^L = H_Z + H_{int} + H_{RF}^L(t)$$

Rotující systém

$$H^R = H_{int} + H_{RF}(t)$$

frekvence

časově proměnná amplituda  
(pulsní sekvence)

proměnná pouze amplituda

$$e^{-iH_{RF} t}$$



„Interakční“ systém

$$H^I(t)$$

– studium přímého vlivu  $H_{RF}$  na interní Hamiltonián

průměrný Hamiltonián

$$\overline{H} = \frac{1}{T} \int_0^T H^I(t) dt$$

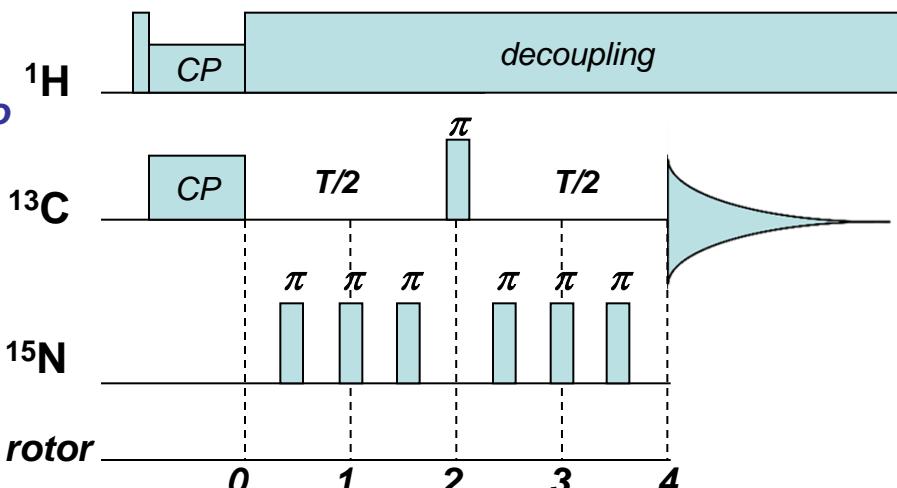
pomocí vhodné pulzní sekvence lze vytvořit **Hamiltonián požadovaného tvaru**

# Heteronuclear dipolar recoupling

**REDOR**

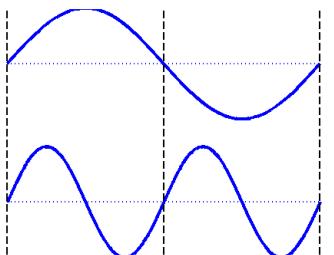
*Rotational Echo*

*Double Resonance*



Principle

$$H_{DD}^I(t) = \sum_{m=-2}^{+2} \omega_{DD}^m \exp(im\omega_R t) I_z S_z$$



**no  $\pi$ -pulses**

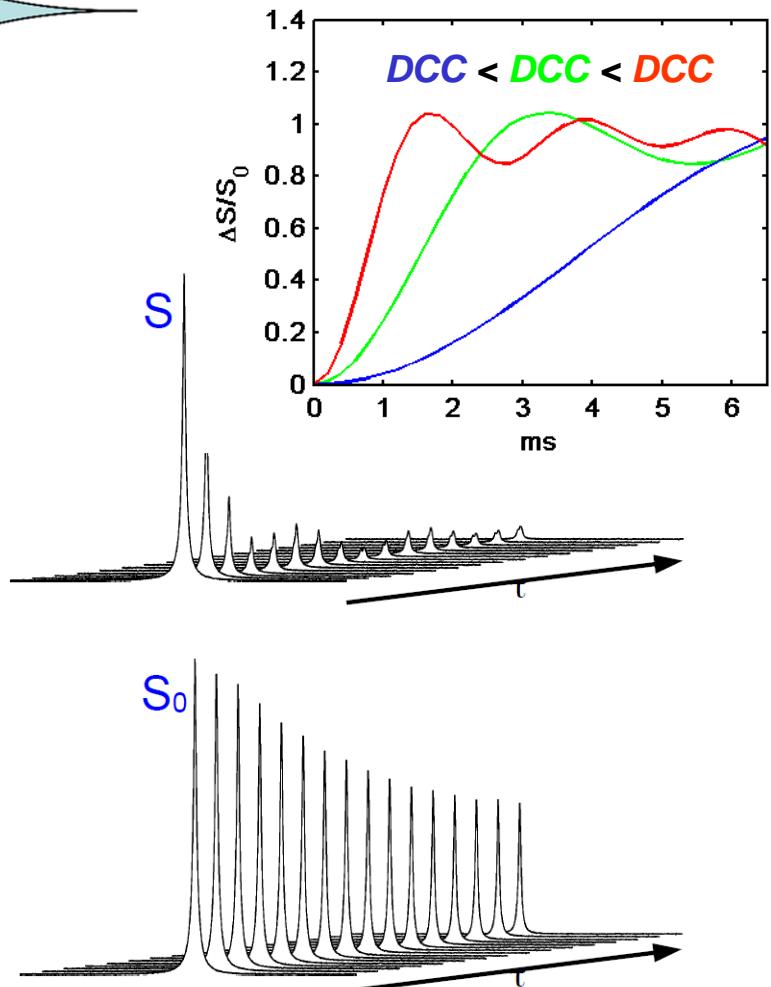
$m = \pm 1$

$m = \pm 2$

**with  $\pi$ -pulses every half of rotor period**

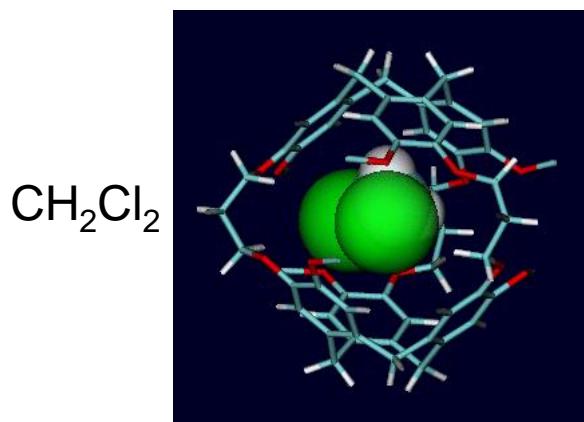
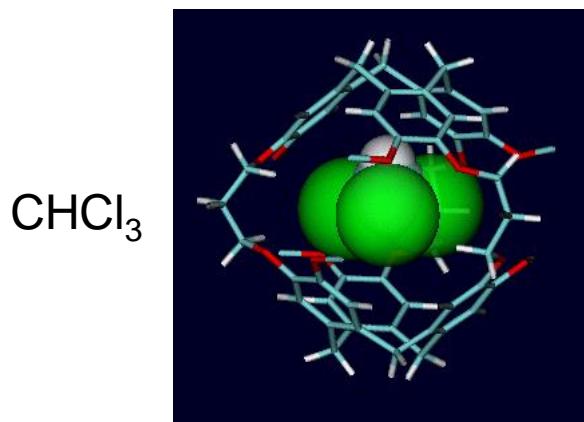
Purpose

- Measure internuclear distances
- Assess motional averaging



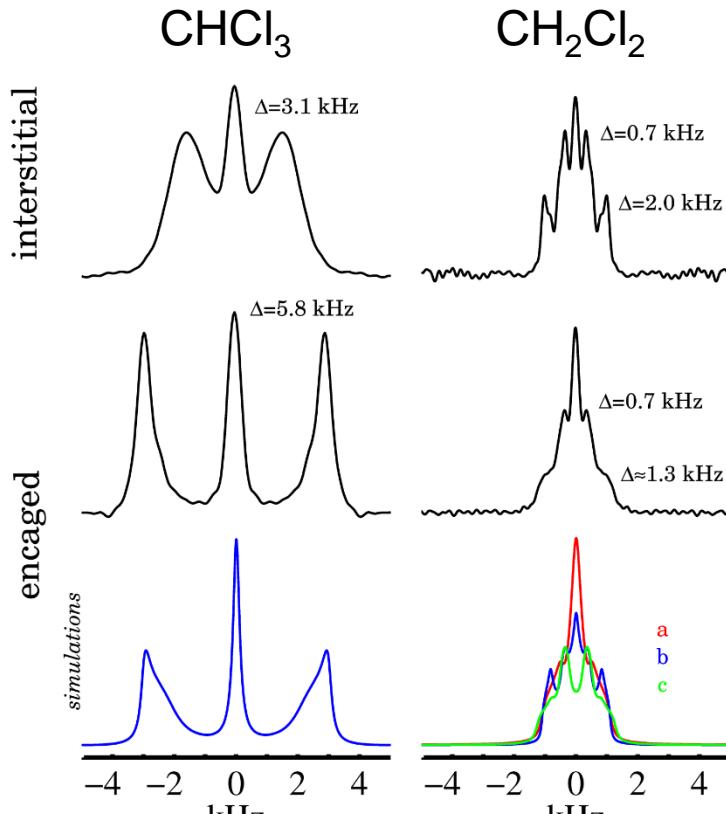
# *Dynamika supramolekulárních komplexů*

# Kryptofan-E



## ***DD interakce vystředovaná pohybem***

## Statická hodnota 21 kHz

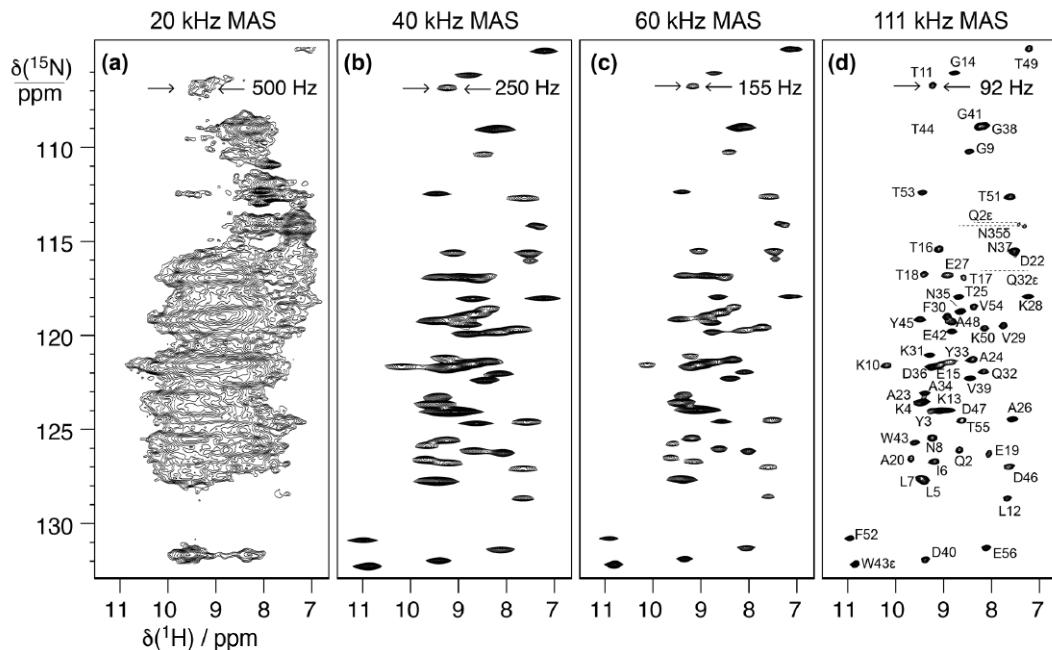
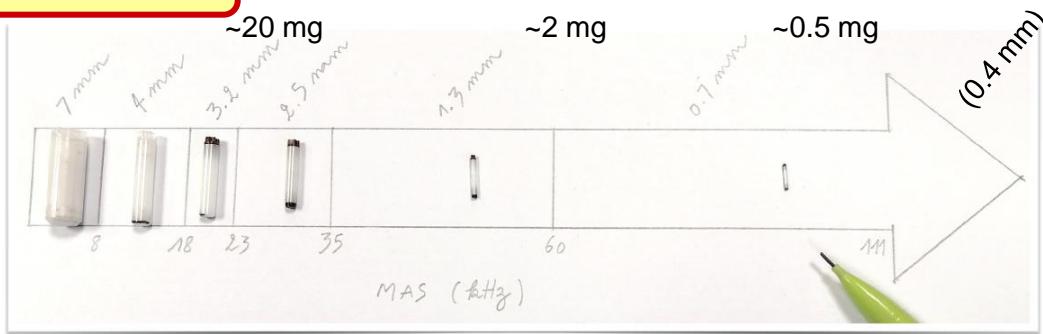


**<DCC> = 19 kHz**

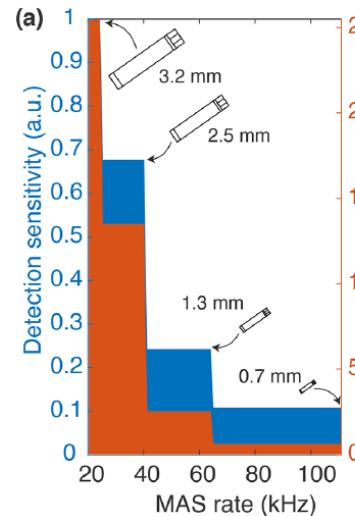
3–5 kHz

# Biological solid-state NMR

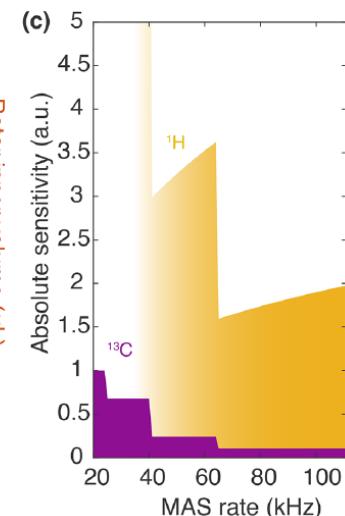
## Faster MAS



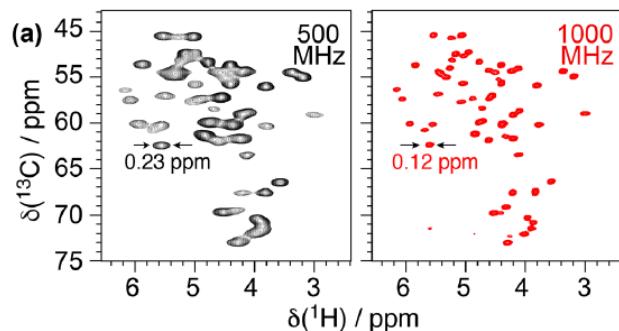
## Advantage of smaller coils



## Advantage of ¹H detection



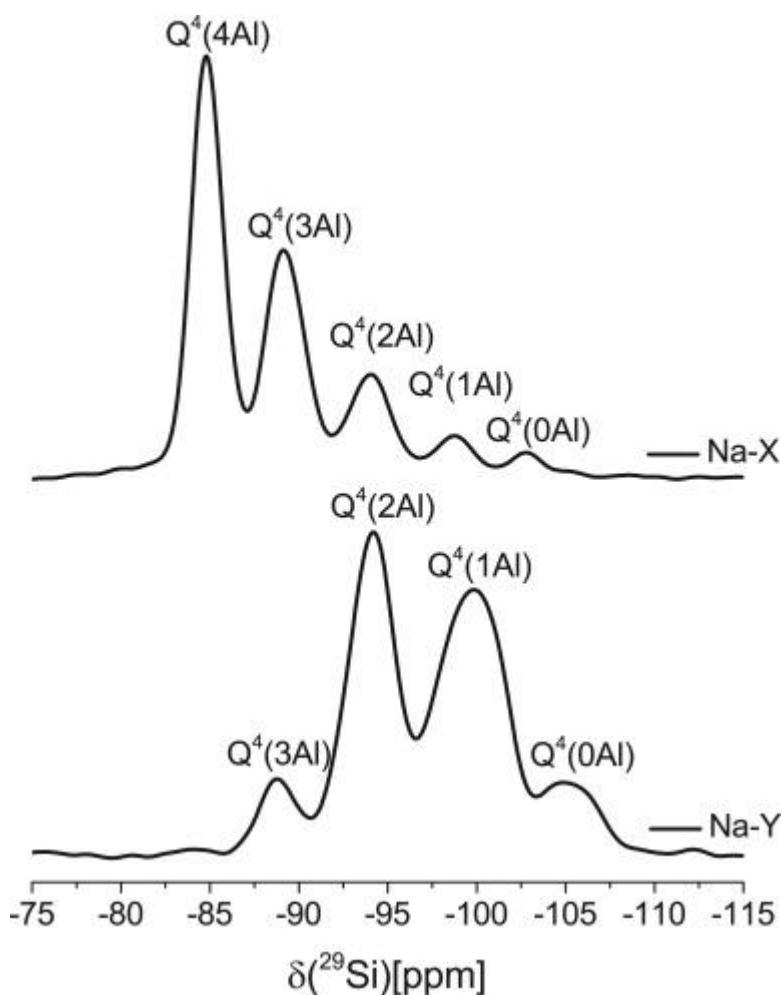
## Higher $B_0$



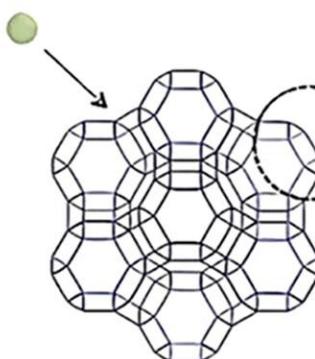
# NMR spectra of zeolites

## $^{29}\text{Si}$ MAS

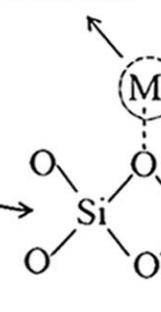
Na-X (top, Si/Al = 1.3) and Na-Y (bottom, Si/Al = 2.7).



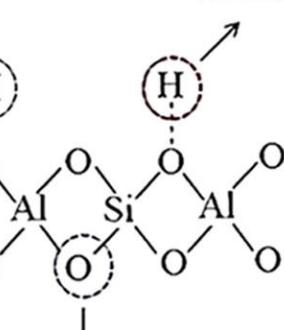
Metal ion



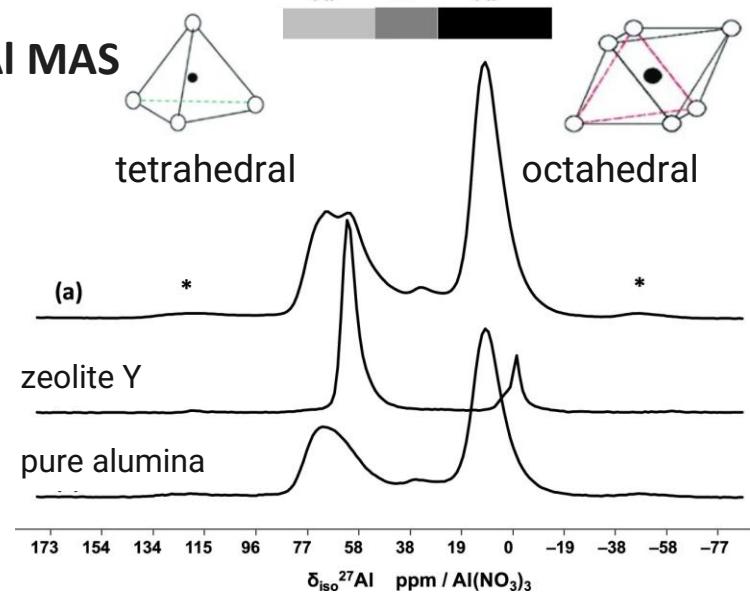
L acid site



B acid site



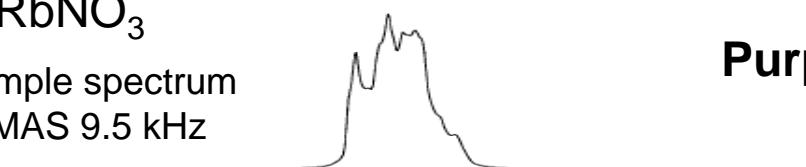
## $^{27}\text{Al}$ MAS



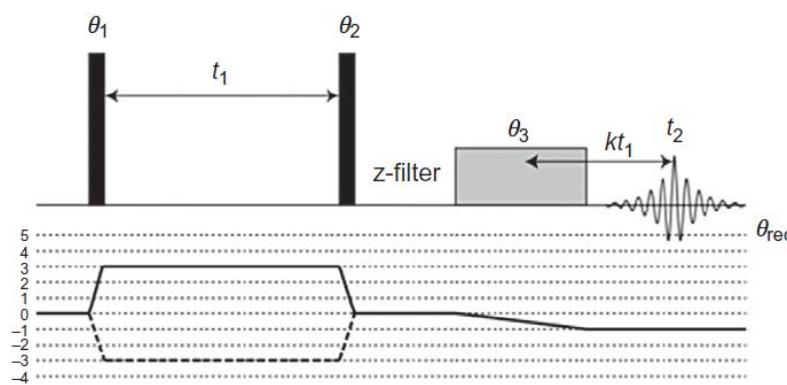
# MQ MAS experiment

$^{87}\text{RbNO}_3$

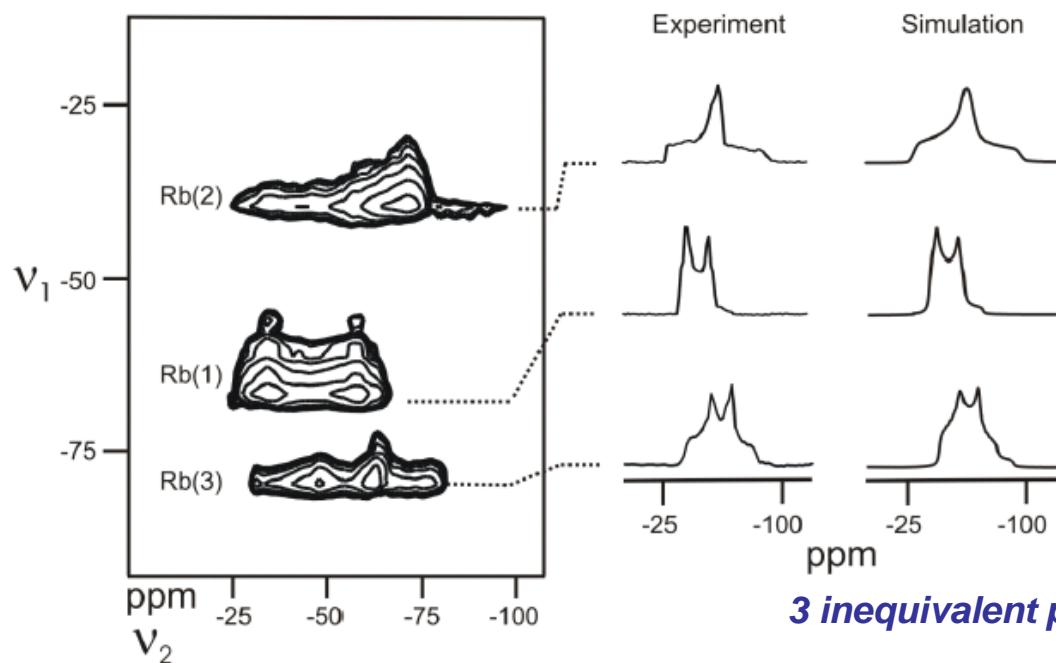
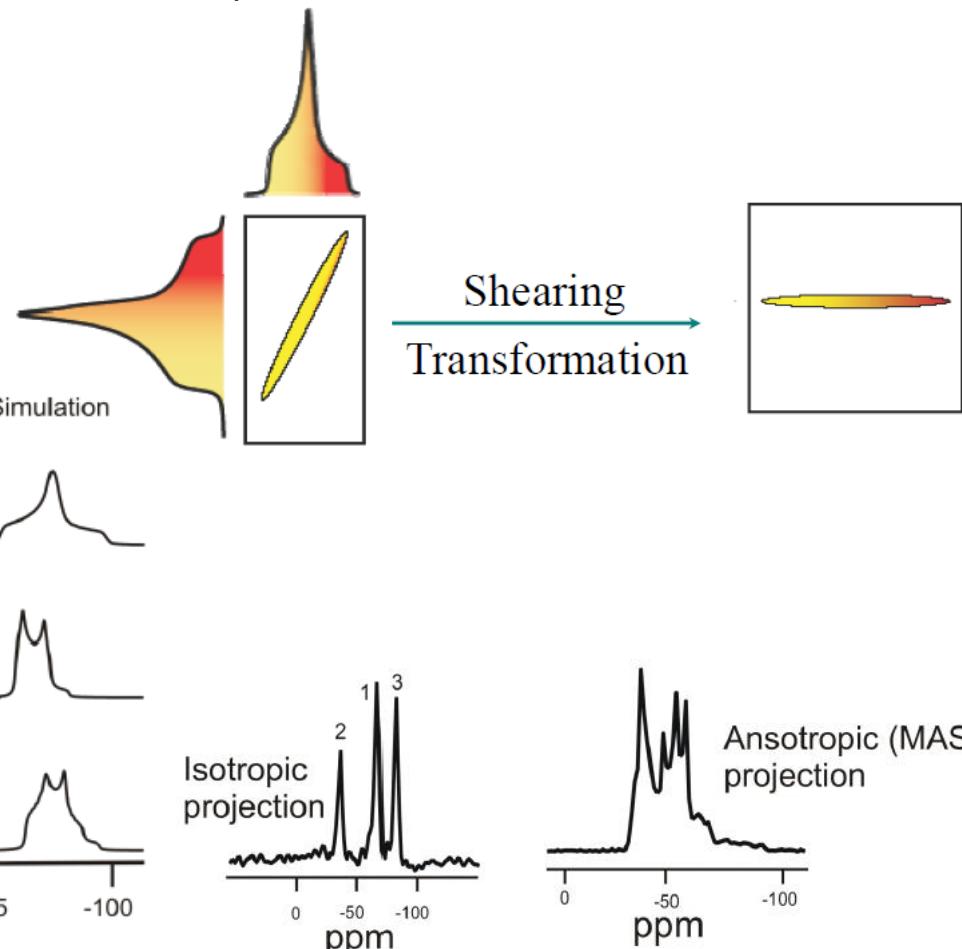
Simple spectrum  
MAS 9.5 kHz



**Purpose** recover high resolution by removing effect of quadrupolar broadening



2D experiment that correlates MQ with SQ

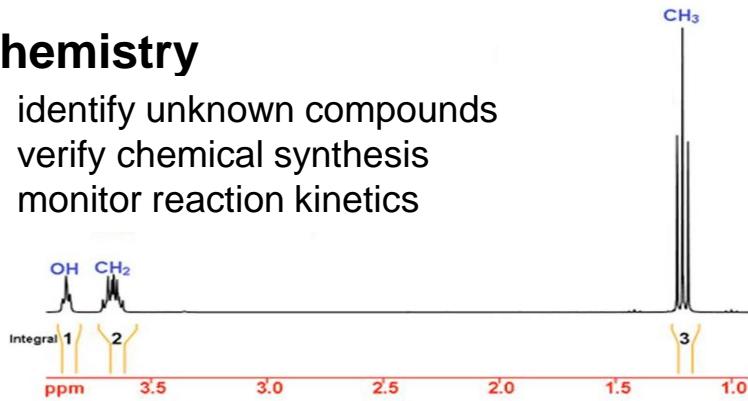


*3 inequivalent positions of Rb atom*

# Nuclear Magnetic Resonance

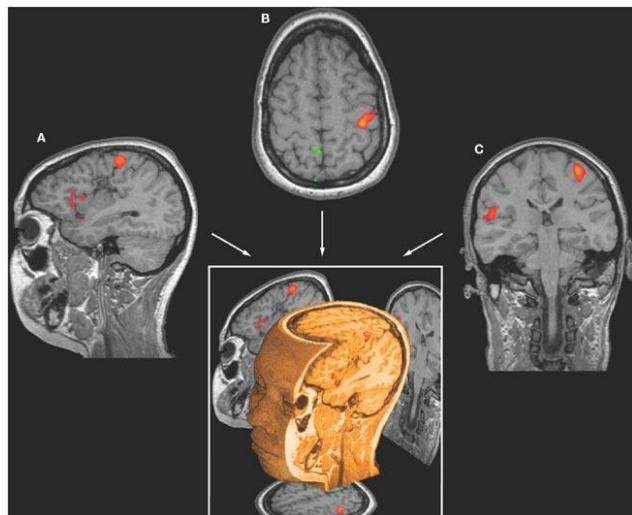
## Chemistry

- identify unknown compounds
- verify chemical synthesis
- monitor reaction kinetics



## Medicine

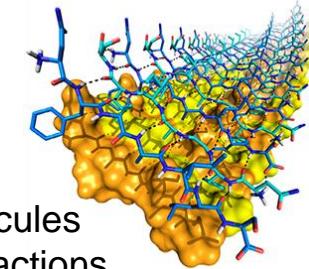
- non-invasive diagnostic imaging
- brain disorders, tumours, and musculoskeletal injuries  
angiography, functional imaging of brain
- metabolomics (diagnostics and treatment monitoring)



Right hand movement

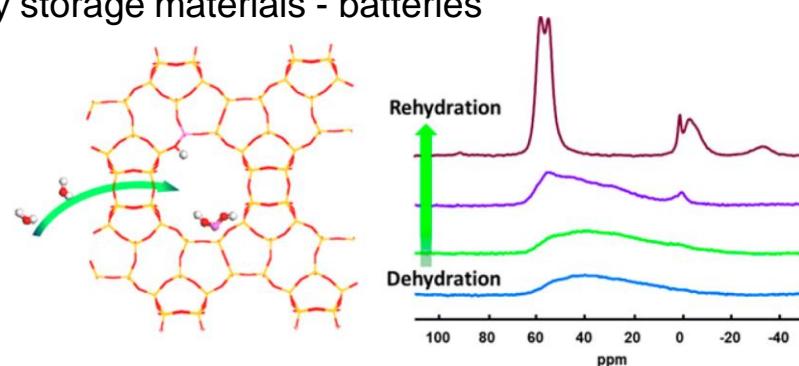
## Structural biology

- determine 3D structures of biomolecules
- insights into their function and interactions
- drug design



## Material science

- polymers, nanomaterials, amorphous materials
- Characterisation of functional materials - catalysts
- energy storage materials - batteries



## Environmental science

- Pollutants (aerosols)
- soil composition (organic matter)

## Pharmaceutic industry

- Drug formulations, quality control
- Polymorphs and bioavailability