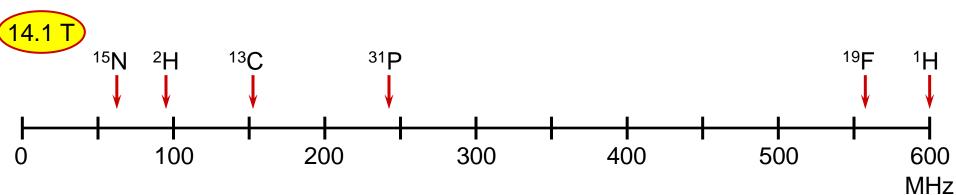
Methods of Double resonance

¹³C spectra and heteronuclear decoupling, experiments APT and DEPT

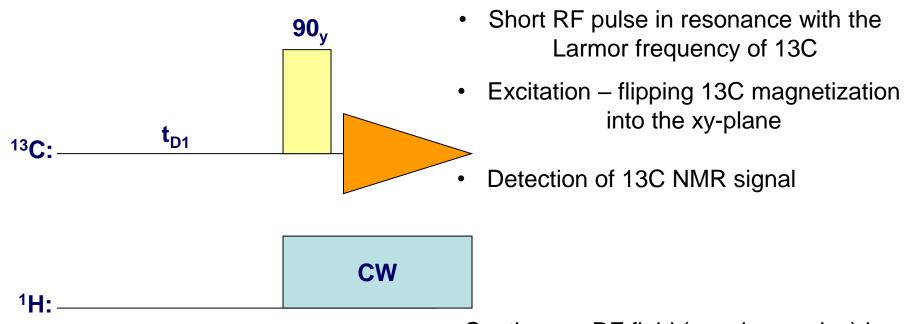
Double resonance



- The Larmor frequencies of the different isotopes are very different
- The radiofrequency field is in resonance with only one isotope
- Multiple radiofrequency irradiations with different frequencies can be used at the same time

Double resonance- two RF fields in resonance with two isotopes

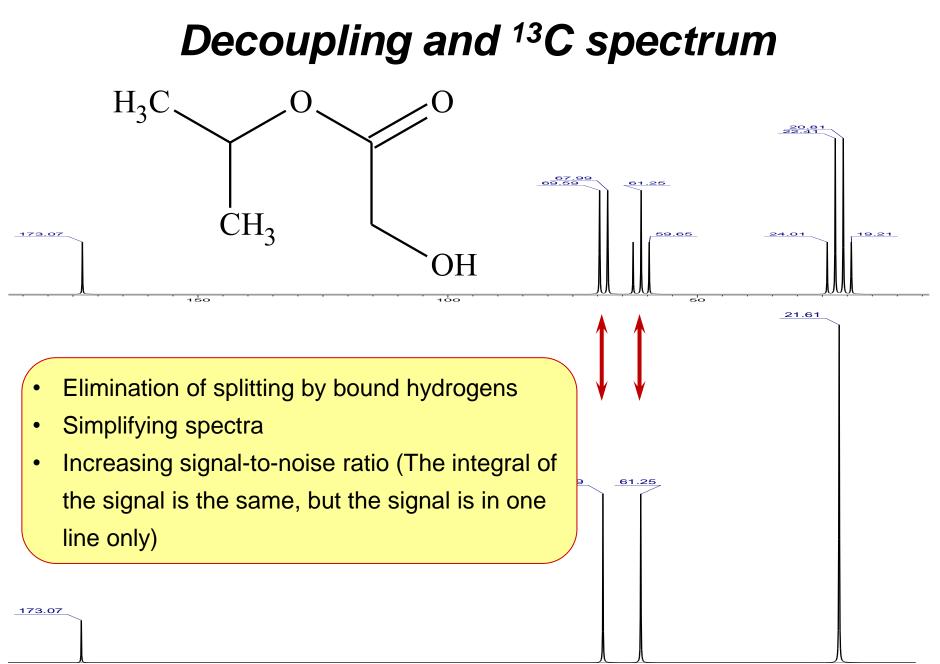
Decoupling and ¹³C spectrum



 Continuous RF field (very long pulse) in resonance with the Larmor frequency of 1H

 Different irradiation schemes (sequences of different pulses) for higher efficiency of averaging the spin states – CW, WALTZ, GARP, … • Fast jumps between levels $m = \pm \frac{1}{2}$ Averaging = effectively spin 0





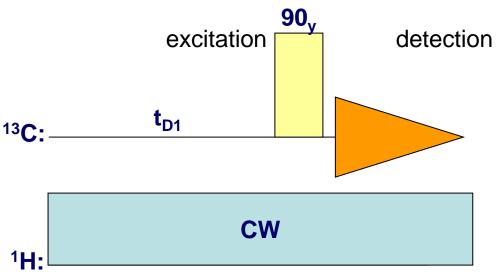
Decoupling and heteronuclear NOE

 13C magnetization is along the z-axis (before the excitation pulse)

- Continuous RF field in resonance with 1H Larmor frequency
- saturation equalization of level occupancies (of 1H transitions)
- Influencing intensity in 13C transitions = Nuclear Overhauser effect (NOE)

NOE increases signal strength of 13C nuclei with directly bonded hydrogens (up to by 199%) Depends on relaxation - local mobility The enhancement is different for different carbons

¹³C experiment with continuous decoupling



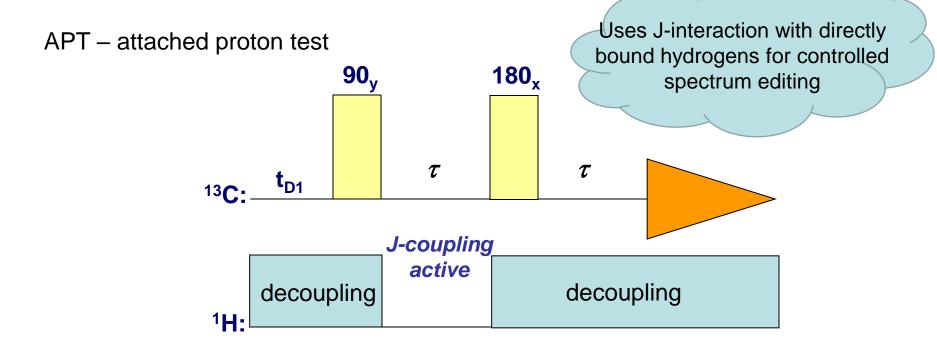
- Natural abundance of 13C is 1%, hence the low signal strength
- The measurement is repeated many times, signal accumulates, noise cancels randomly
- There is a time delay between each measurement when the 13C magnetization recovers (T1 relaxation)

13C spectra are not quantitative • NOE enhancement

Insufficient relaxation

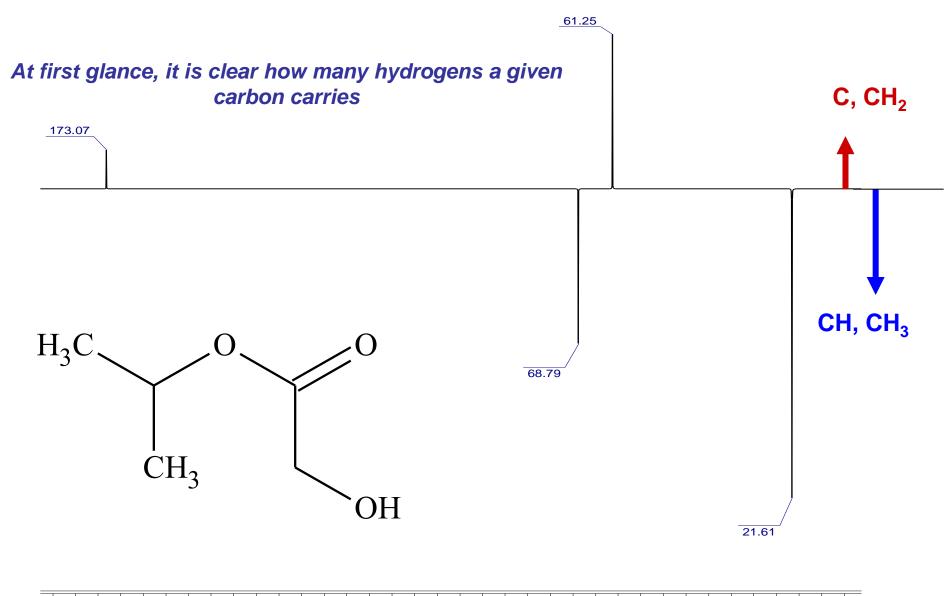
NOTE: measurement can also be set for quantitative response!!!

APT Experiment

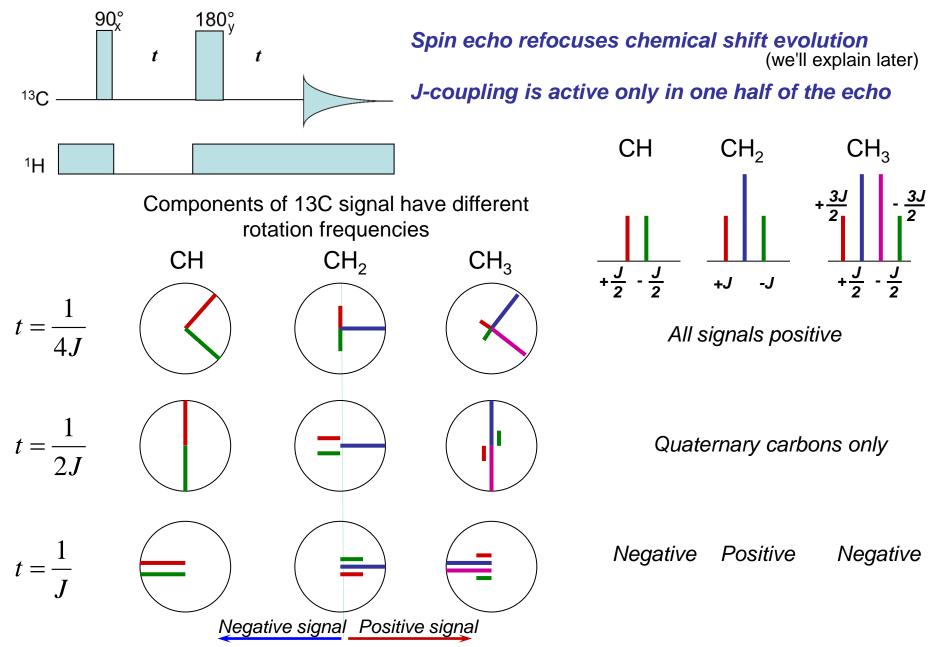


Amplitude of 13C signal is modulated depending on the number +1 positive bound hydrogens +0.5 0 1/2 1/J Pulse sequence -0.5 CH Sophisticated manipulation negative -1 4 with spin system (magnetization) CH₃

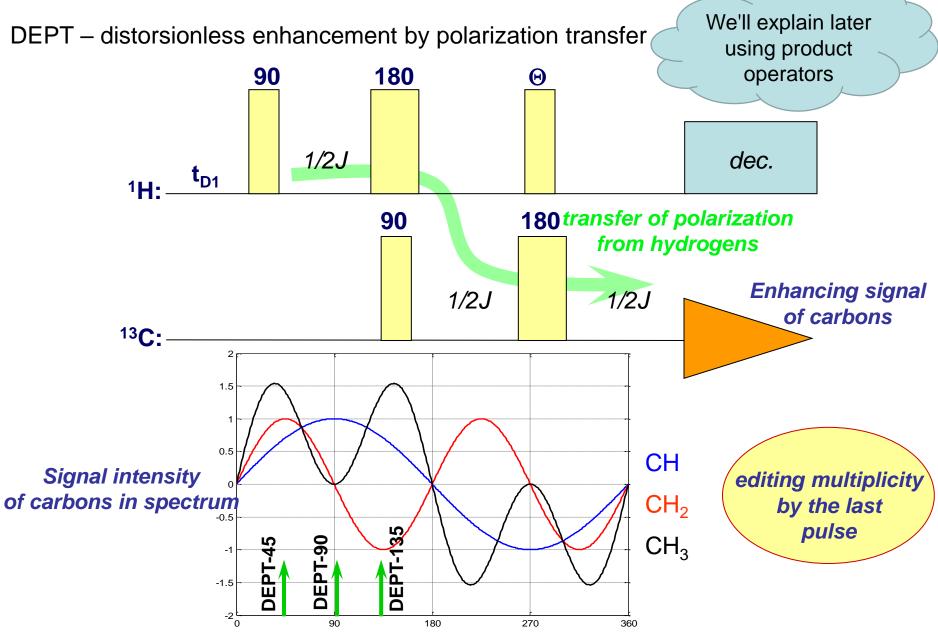
APT experiment



APT experiment – explanation



DEPT experiment



DEPT experiment $\begin{array}{c} O \\ \parallel c \\ \bullet \\ d \\ CH_3 - C - CH - CH_2OH \\ a \\ \parallel \end{array}$ 4-Hydroxy-3-Methyl-2-Butanone CH_3 CH a CH₃ positive е CH₂ negative **DEPT-135** DEPT does not display quaternary carbons only CH DEPT-90 all DEPT-45 protonated carbons APT CDCI3 b d ¹³C Spectrum С е а 220 200 180 160 140 120 100 80 60 **4**0 20 0 ppm 240